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## Evidence from molybdenum isotopes for a transient and localized nature of free aquatic oxygen around 3 Ga.

Kramers, J.D.<sup>1</sup>, Beukes, N.<sup>1</sup>, Gumsley, A.<sup>1</sup>, Guy, B.<sup>2</sup>, Naegler, Th.<sup>3</sup>, Pettke, Th.<sup>3</sup>, Smith, A.B.<sup>1</sup> and Voegelin, A.<sup>3</sup>

<sup>1</sup>University of Johannesburg, Department of Geology, Auckland Park 2006, South Africa; [jkramers@uj.ac.za](mailto:jkramers@uj.ac.za)

<sup>2</sup>Mineral Services, SGS South Africa (Pty) Ltd., Booyens, Johannesburg 2091, South Africa

<sup>3</sup>Institute of Geological Sciences, University of Berne, 3012 Berne, Switzerland

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Evidence of early occurrences of low levels of free molecular oxygen has been reported from banded iron formations in the West Rand Group [1] (lower Witwatersrand Supergroup) as well as at the base of its more distal stratigraphic equivalent [2], the Mozaan Group. Here oxygenated conditions have been documented by Cr isotopes on a paleosol [3] and Mo isotopes in banded iron formation [4]. Mo systematics are a powerful in the study of oxygenation in early Earth history. In organic-rich shales and manganese oxide crusts, the combination of Mo concentrations >2 ppm and  $\delta^{98/95}\text{Mo}$  variations > 0.3‰ signal free oxygen at levels as low as  $10^{-5}$  PAL or less [5], as long as these are present in a sufficiently large reservoir [6]. In order to explore the marine redox conditions at c. 2.95 Ma in a broader stratigraphic and spatial context, we have determined Mo contents and  $\delta^{98/95}\text{Mo}$  values of 20 (mainly carbonaceous) mudrock, banded iron formation (BIF) and manganese concretion samples throughout the West Rand Group as well as 3 samples from the Booyens Formation, Central Rand Group. Further, we analysed 8 similar samples from the Ntombe, Delfkom and Bangaspoort Formations in the Mozaan Group. Mn concretions occur as  $\text{Mn}^{2+}$  carbonate, but reflect original precipitation as  $\text{MnO}_2$  [7], which requires some free  $\text{O}_2$  even if bacterially mediated [8].

In mudrocks throughout the West Rand Group, we found Mo concentrations varying from 0.2 to 3 ppm and  $\delta^{98/95}\text{Mo}$  values from 0.1 to 0.45‰. Mostly, higher concentrations coincide with higher  $\delta^{98/95}\text{Mo}$  values. A carbonaceous mudrock sample from the Booyens Formation has 8 ppm Mo and  $\delta^{98/95}\text{Mo}$  of 0.6‰. There is thus a weak but persistent signal that free  $\text{O}_2$  affected the Mo isotopic mass balance in the shelf environment during West Rand Group times (2970-2910 Ma), and in the overlying Booyens Formation this signal is stronger. Three BIF samples from the lowermost West Rand Group and the Booyens Formation have <2ppm Mo but  $\delta^{98/95}\text{Mo}$  from -0.1 to +0.5‰. However, two Mn-concretions from the Brixton Formation have ~1 ppm Mo and  $\delta^{98/95}\text{Mo}$  of 0.2‰. The samples from the Mozaan Group (mudrocks and Mn concretions alike) present Mo concentrations varying only between 0.2 and 1.5 ppm, with  $\delta^{98/95}\text{Mo}$  between 0.2 and 0.4‰, within the detrital range and thus not indicating free  $\text{O}_2$ . The Mn concretion data present an apparent paradox, since free  $\text{O}_2$  is required for their formation [8]. However, this requirement is local only, whereas oxygenation on a basin-wide scale would be required for dissolved Mo to reach more distal sedimentation sites. We conclude that free  $\text{O}_2$  was confined to 'oxygen oases' [4]. Further, an episode of strongly oxidizing conditions found at the base of the Mozaan Group [3,4] appears to have been transient.

### References:

- [1] Smith AJB, Beukes NJ and Gutzmer J. (2013) *Econ Geol* 108: 111-134.
- [2] Beukes NJ and Cairncross B (1991) *S Afr J Geol* 94: 44-69.

- [3] Crowe SA et al. (2013) *Nature* 501: 535-538.
- [4] Planavsky NJ et al. (2014) *Nature Geosc* 7: 283-286.
- [5] Siebert C et al. (2005) *Geochim Cosmochim Acta*, 69: 1787-1801.
- [6] Wille M et al. (2007) *Geochim Cosmochim Acta* 71: 2417-2435.
- [7] Smith AJB et al. (2014) *Goldschmidt Conf 2014 Abstracts*: 2332.
- [8] Tebo BM et al. (2005) *Trends Microbiol* 13: 421-428.

