

Desalination as a Source of Fresh Water

Desalination 101

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Presentation Overview

- Source water requiring desalination
- Properties of water and their impact on desalination
- Desalination technologies
- Water recovery and cost of desalination
- Issues associated with desalination
 - Pre-treatment of source water
 - Energy demand and energy recovery
 - Concentrate management
- Future technologies

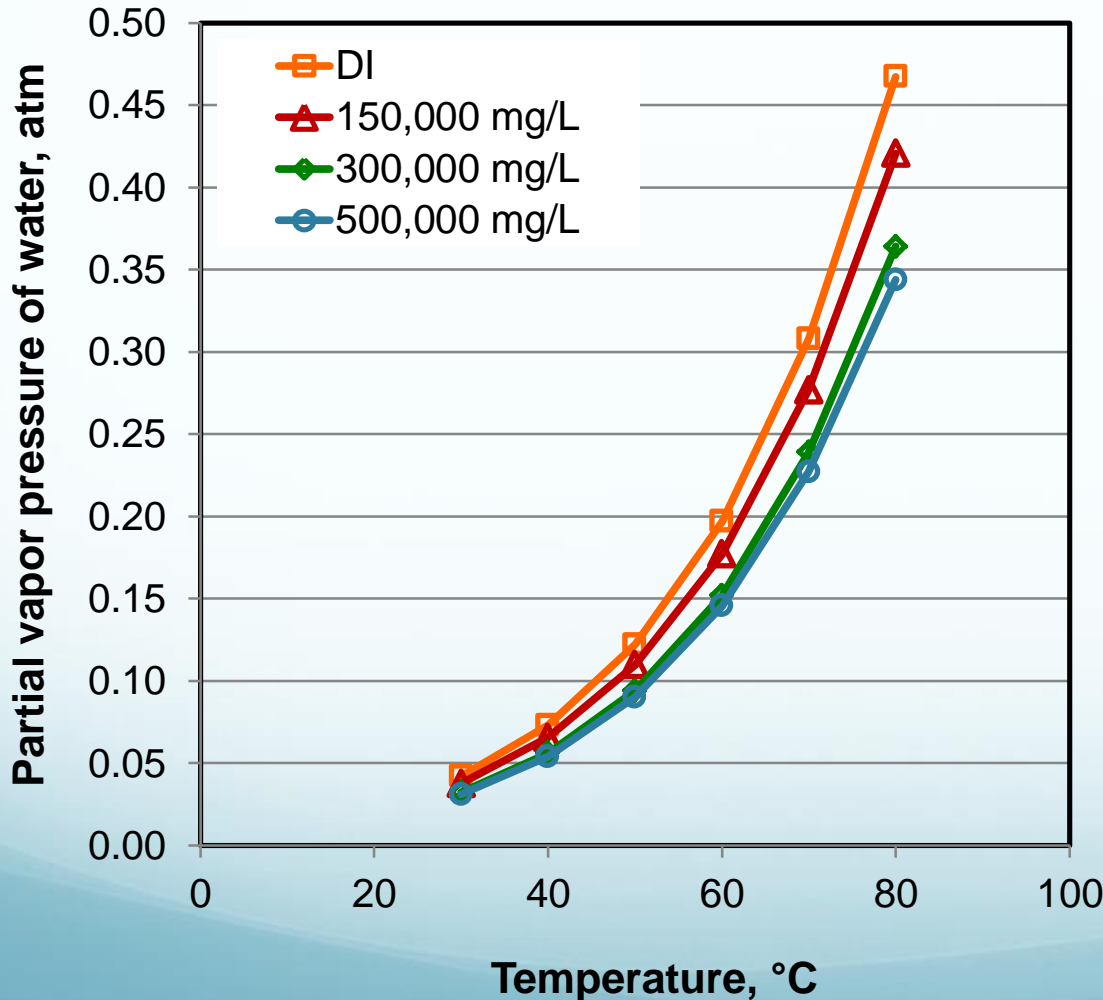
Desalination of Impaired Water

- Common water sources requiring desalination
 - Seawater
 - Brackish groundwater
 - Industrial water (high purity)
 - Domestic and industrial wastewater for advanced reuse
 - Surface water containing specific dissolved solids
- Constituents requiring removal by desalination
 - Dissolved solids, simple salts/ions, heavy metals, nutrients, hardness, organic contaminant of emerging concern, etc.

Colligative Properties of Ionic Solutions

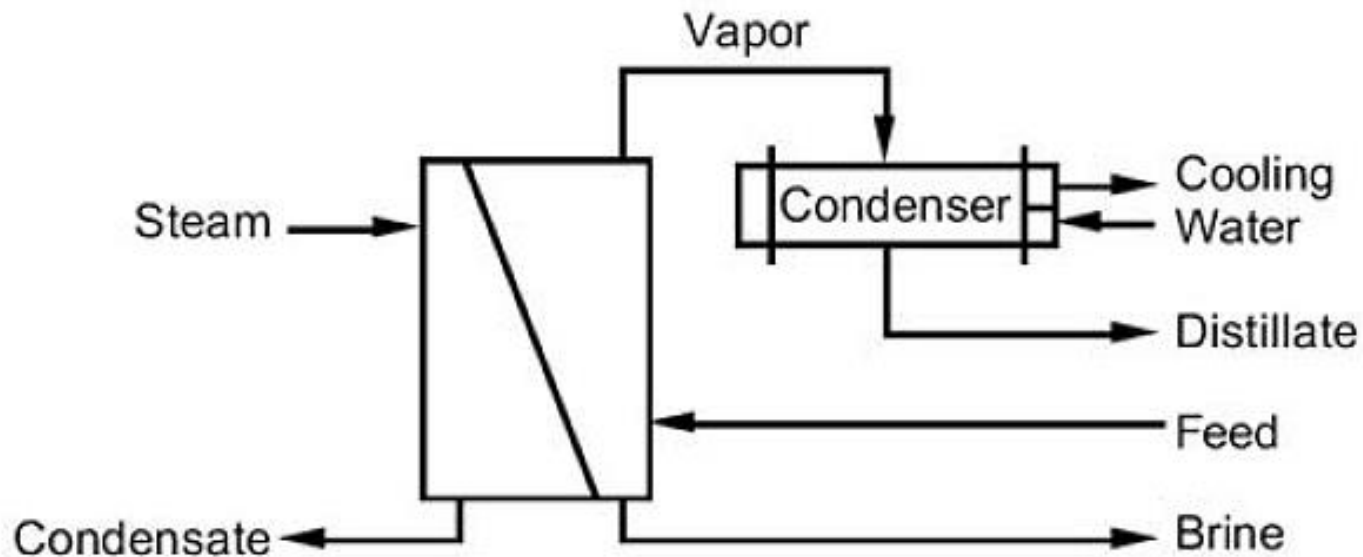
- Properties of solution that depend on the number of solute molecules present, but not on the nature of the solute
- Osmotic pressure, vapor pressure, freezing point depression, and boiling point elevation are examples of colligative properties
- **Osmotic pressure** and **vapor pressure** are two properties of solution that play a major role in desalination

Vapor Pressure of Water

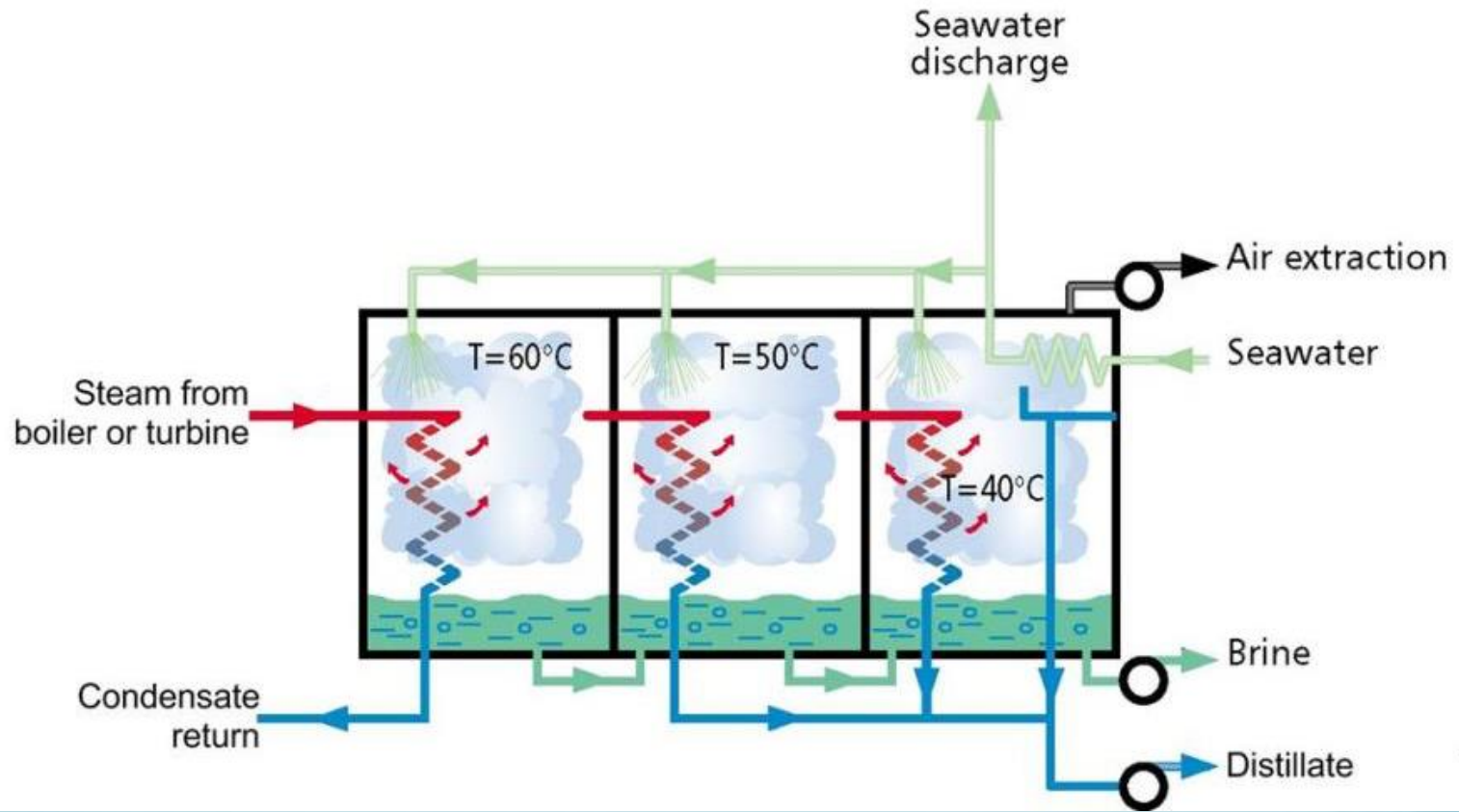


- Increases with increasing temperature
- Minimally decreases with increasing concentration

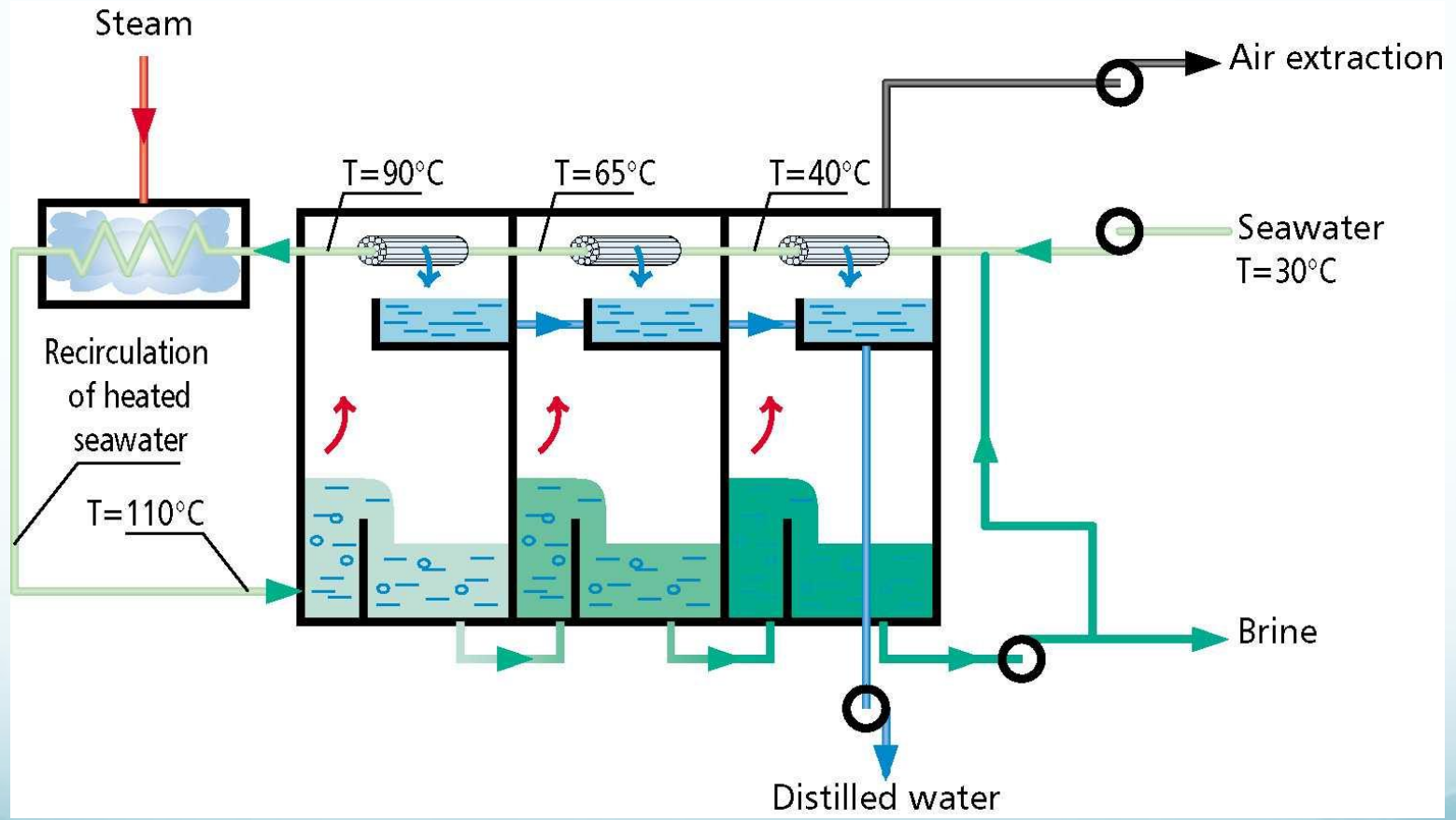
Single-effect Evaporator



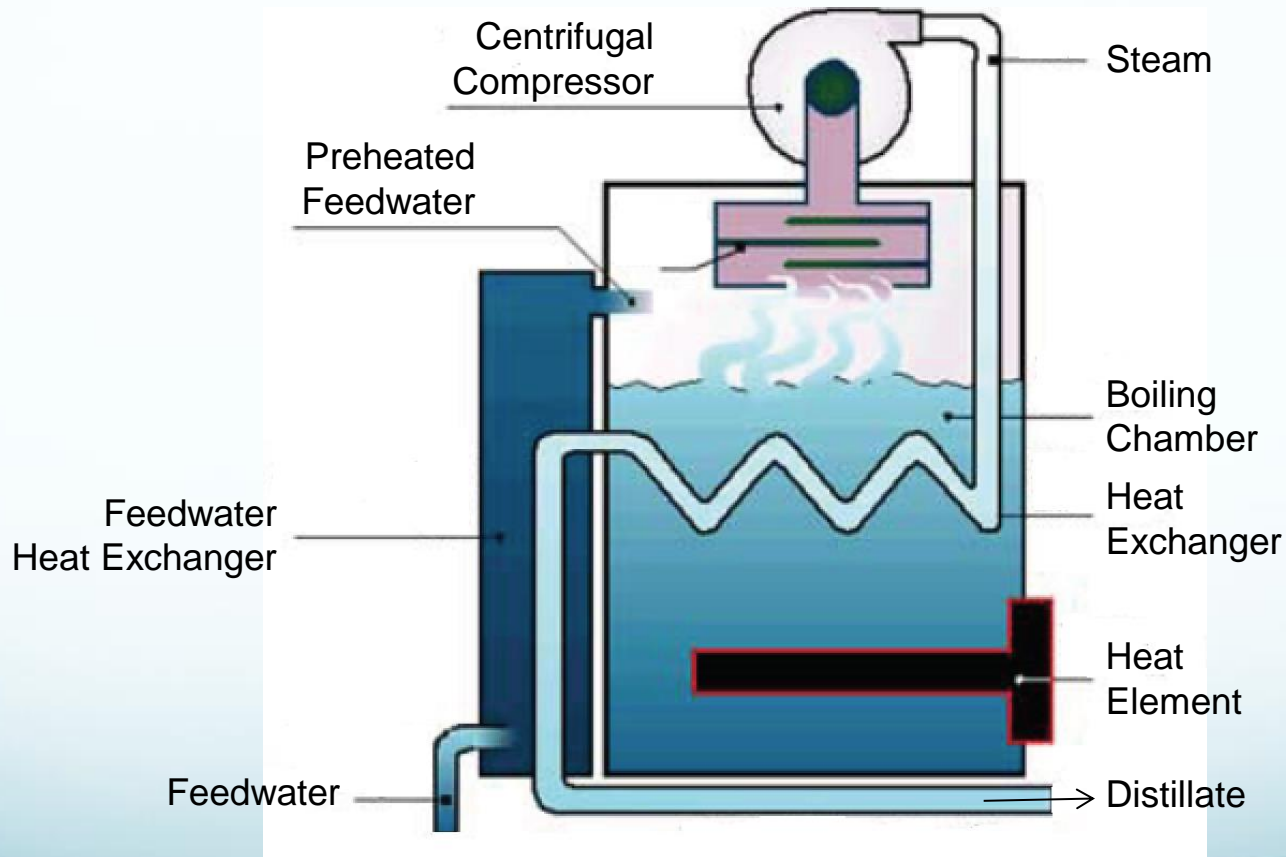
Multiple-effect Evaporator (MED)



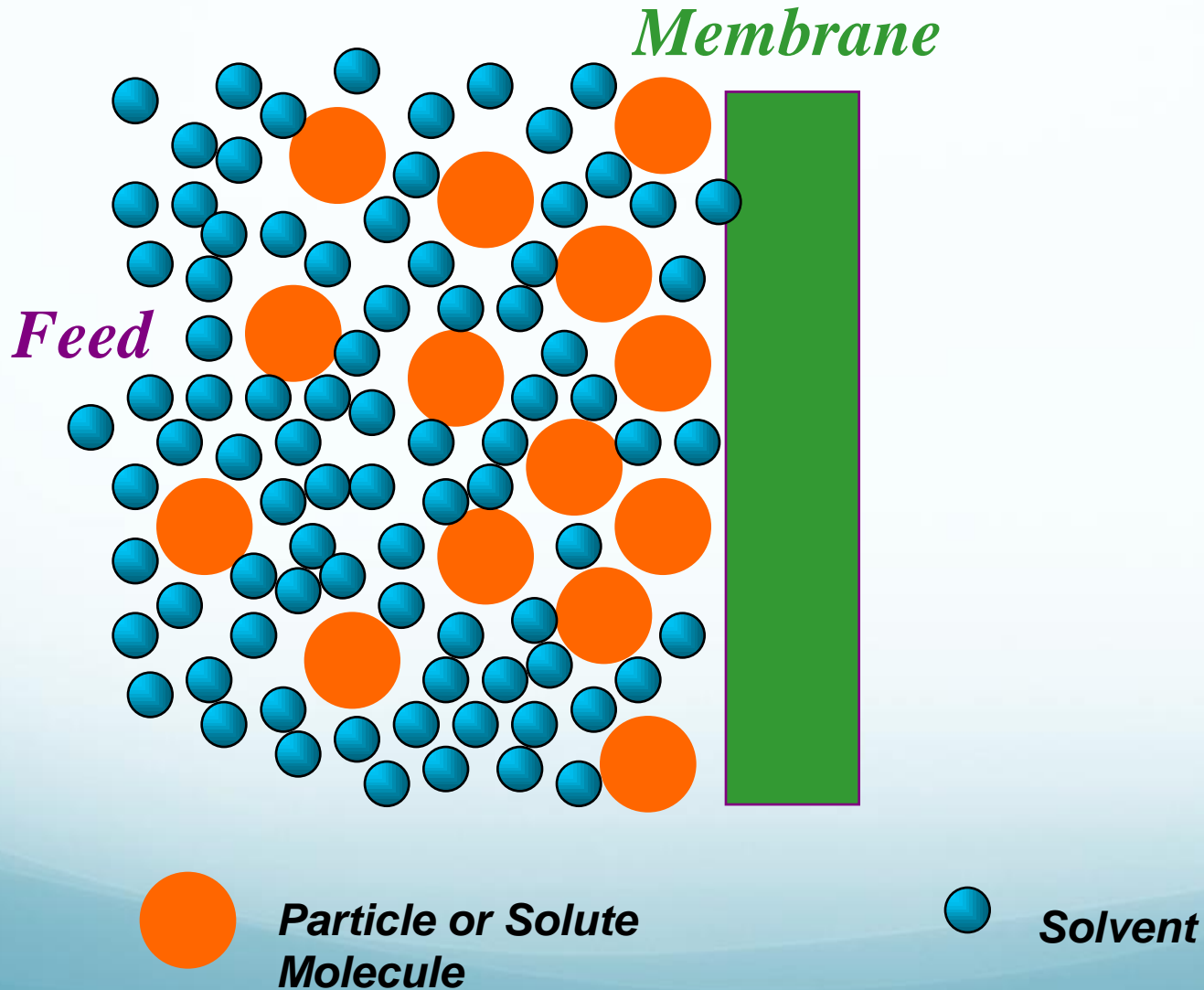
Multistage Flash Evaporation (MSF)



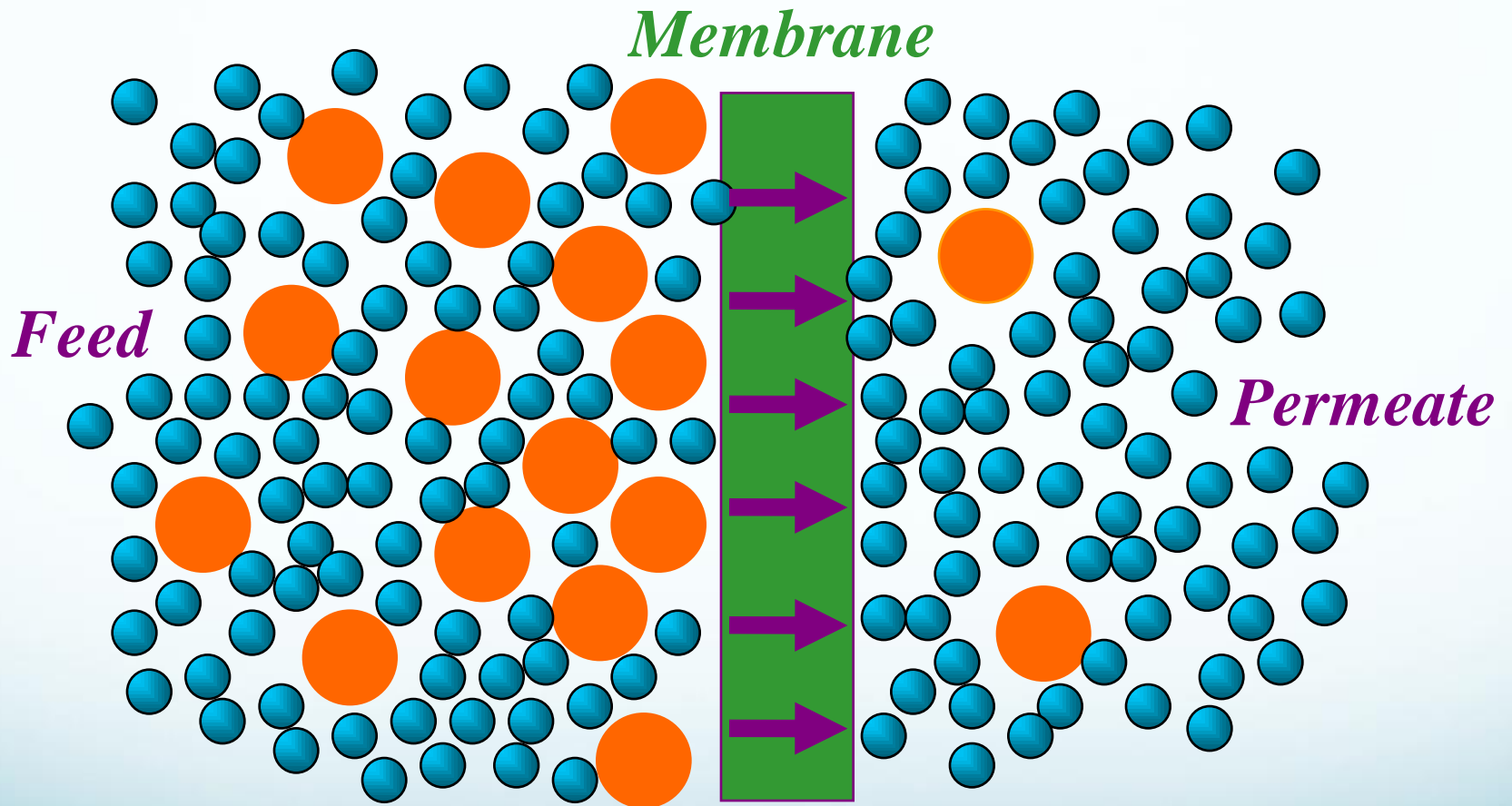
Vapour Compression Brine Concentrator



Membrane Separation



Membrane Separation



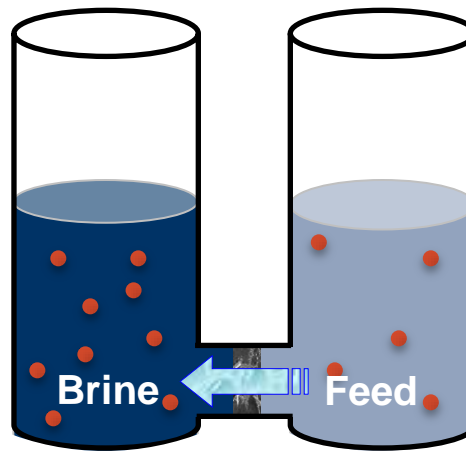
**Particle or Solute
Molecule**



Solvent

Osmotic Pressure of Solution

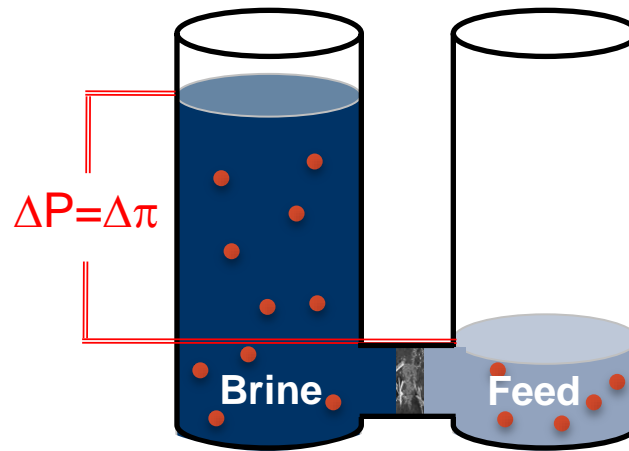
- Increases with increasing solute concentration
- Increases with increasing temperature



Osmosis

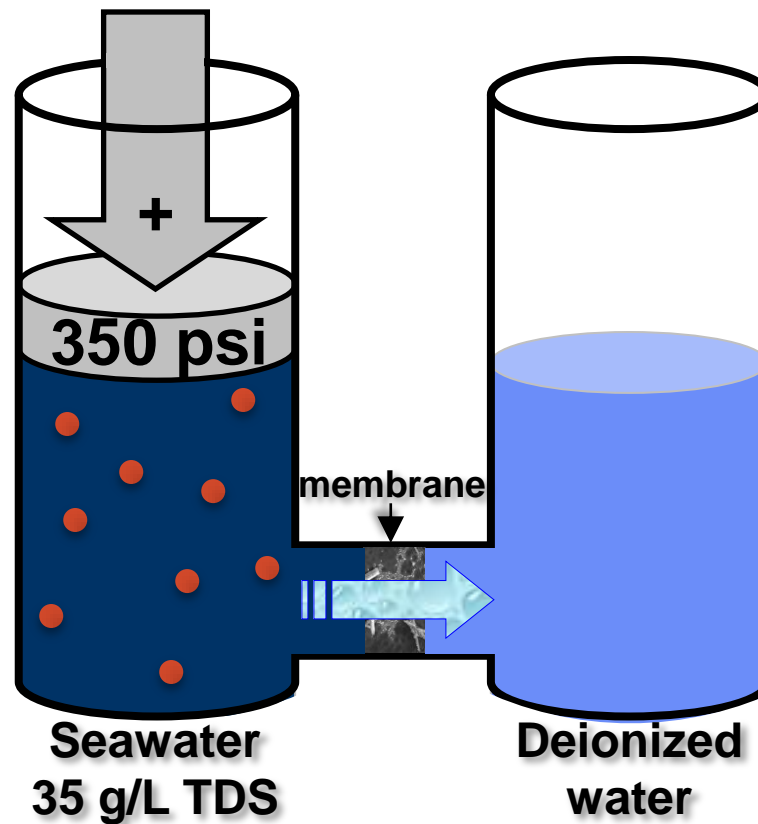
Osmotic Pressure of Solution

- Increases with increasing solute concentration
- Increases with increasing temperature

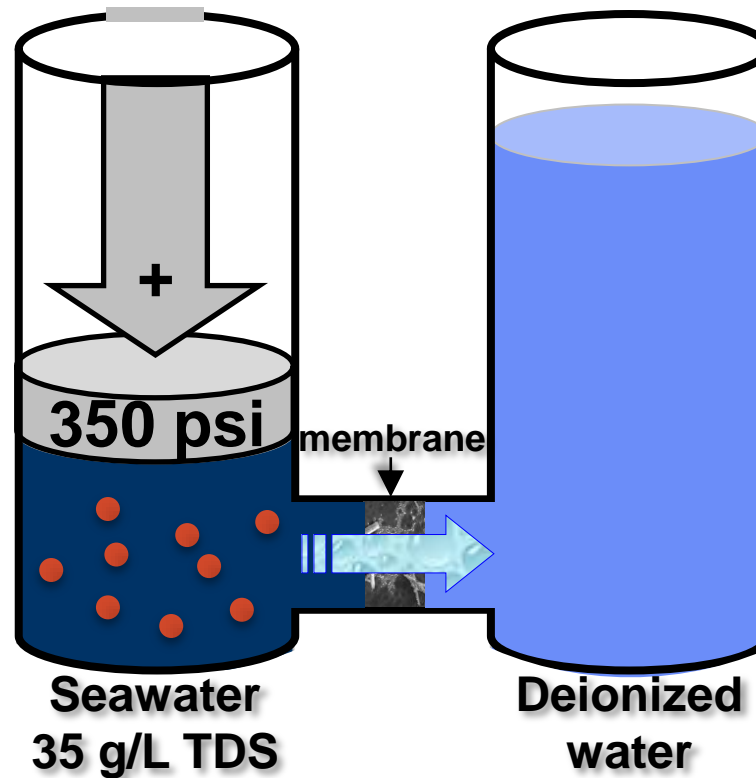


Osmosis

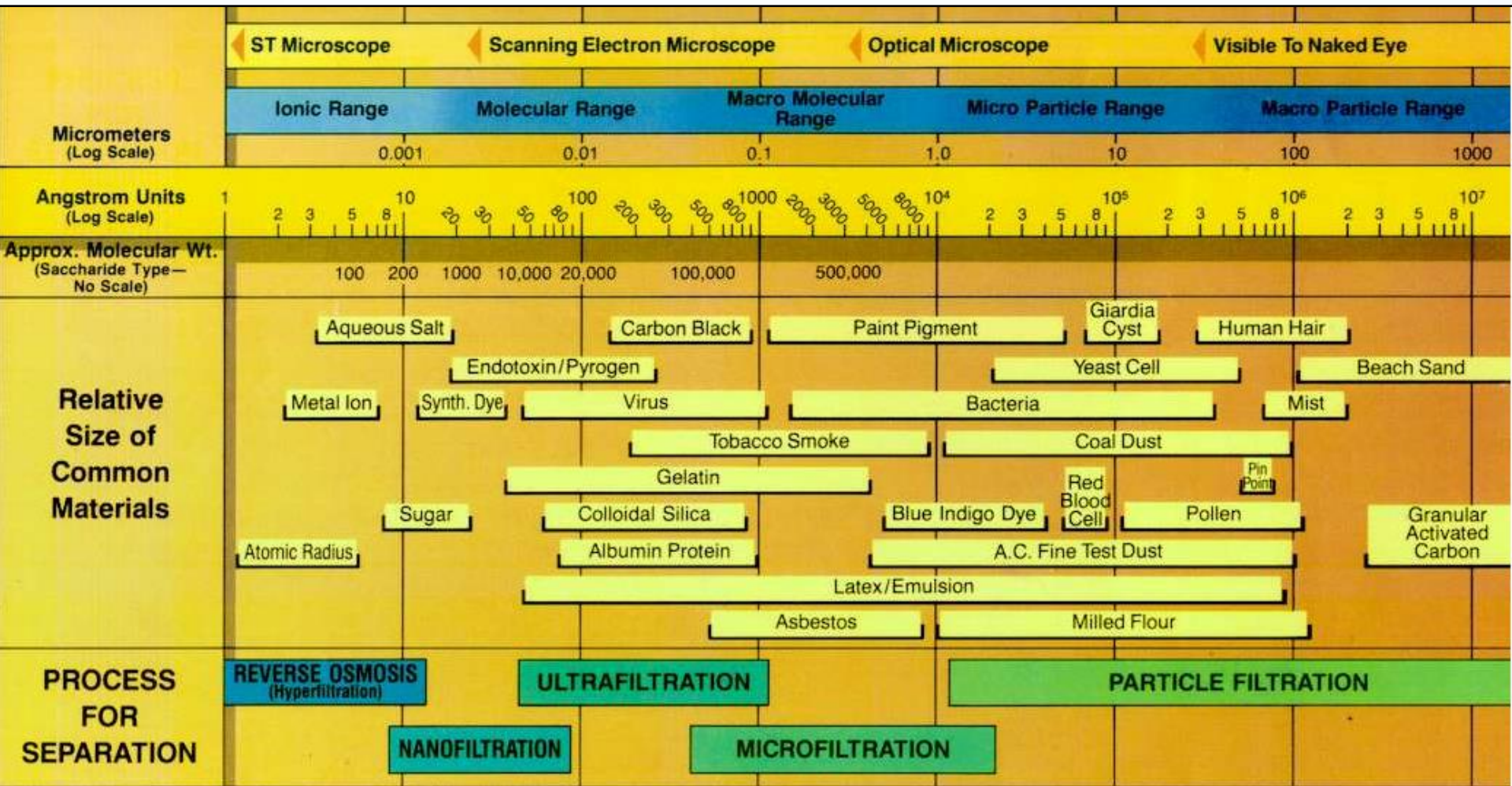
Overcoming Osmotic Pressure



Overcoming Osmotic Pressure



Domains of Filtration



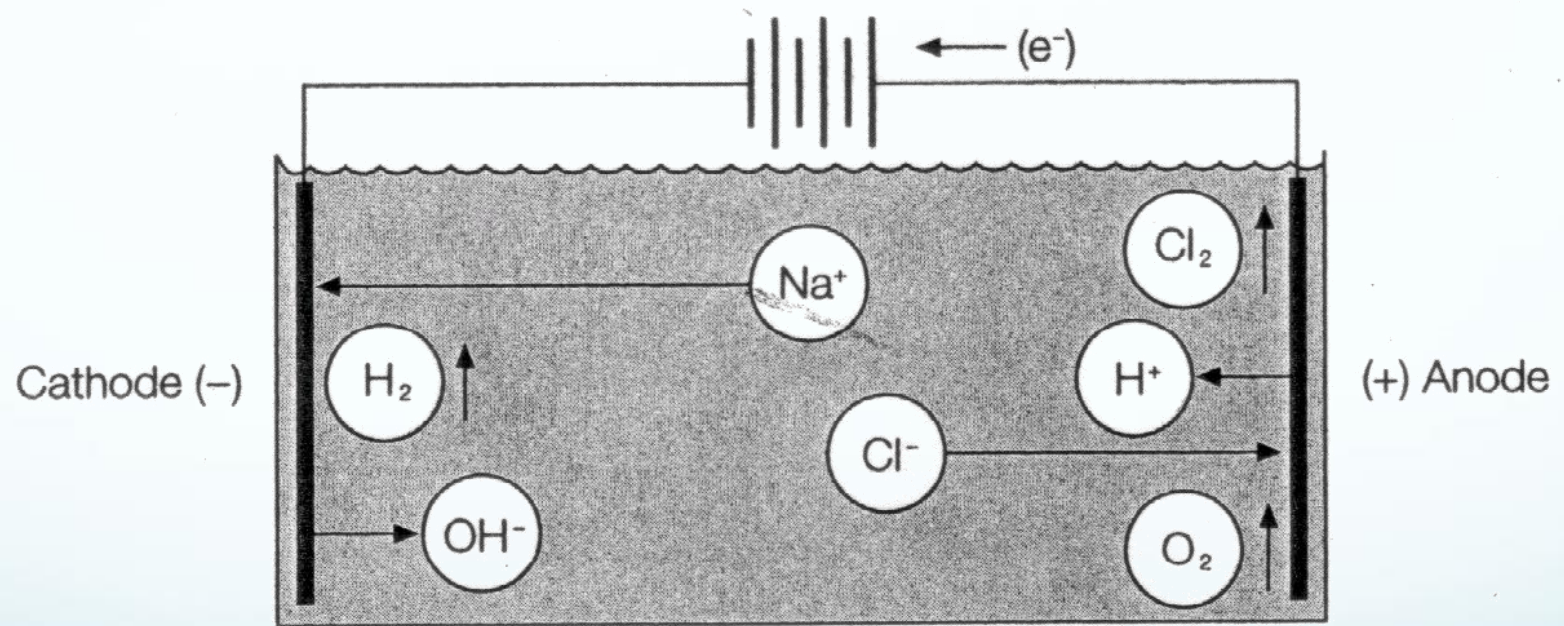
Note: 1 Micron (1×10^{-6} Meters) = 4×10^{-5} Inches (0.00004 Inches)
 1 Angstrom Unit = 10^{-10} Meters = 10^{-4} Micrometers (Microns)

Ranges of Pressure and Flux

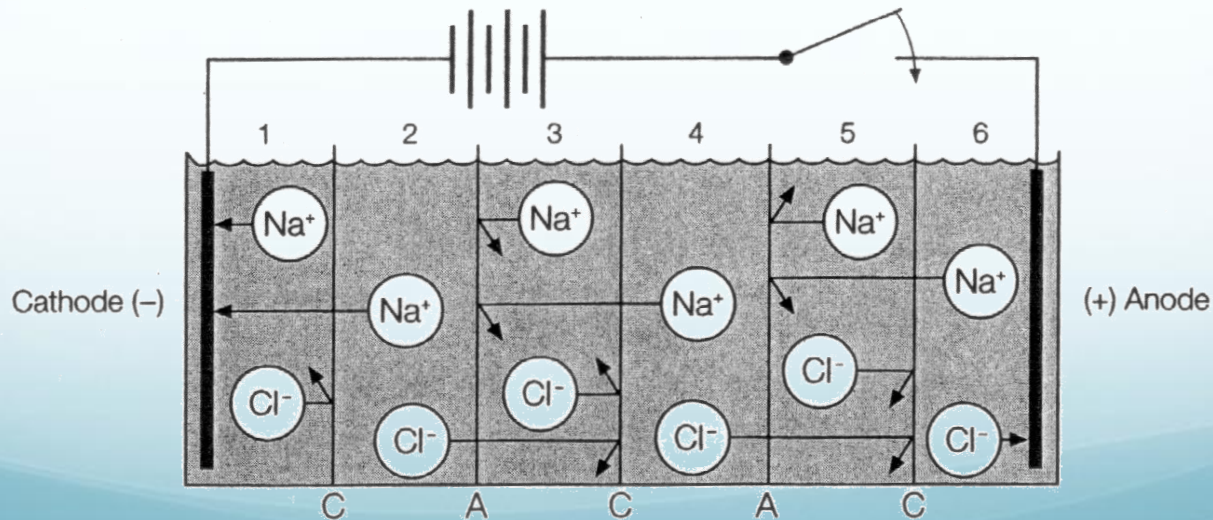
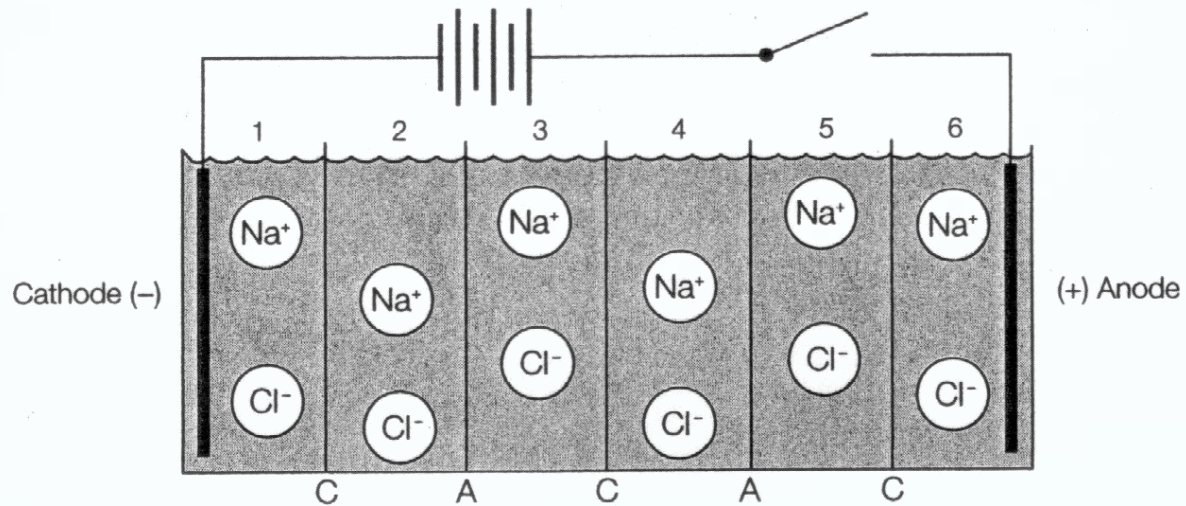
Process	Pressure		Water Flux	
	PSI	kPa	gal/ft ² /day (GFD)	L/m ² /hr (LMH)
Nanofiltration	100 – 400	700 – 2800	15 – 30	20 – 50
Reverse Osmosis	200 – 1000	1400 – 7000	15 – 30	20 – 50

GFD = LMH x 1.7

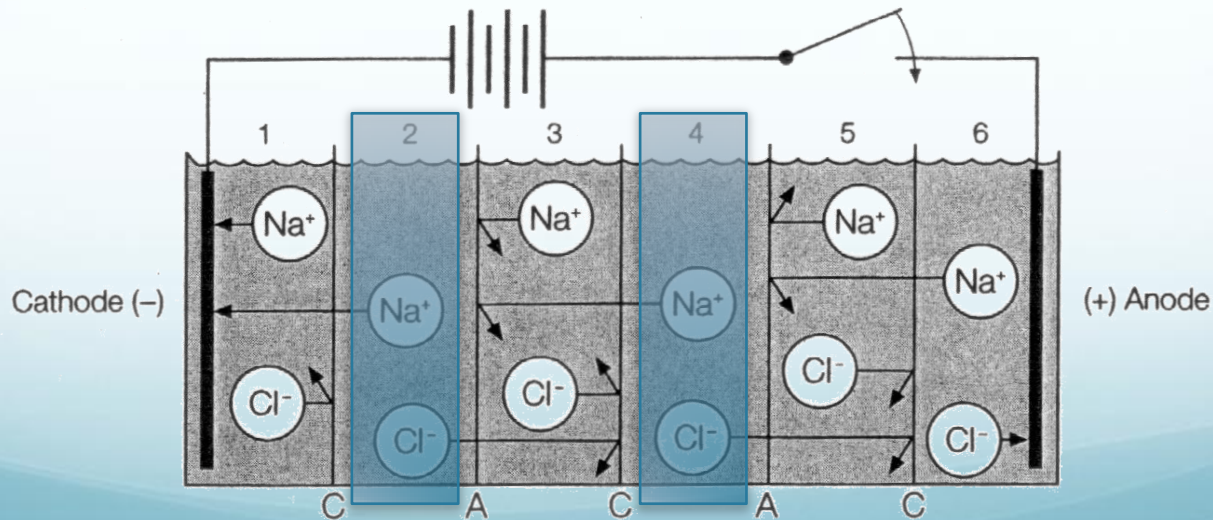
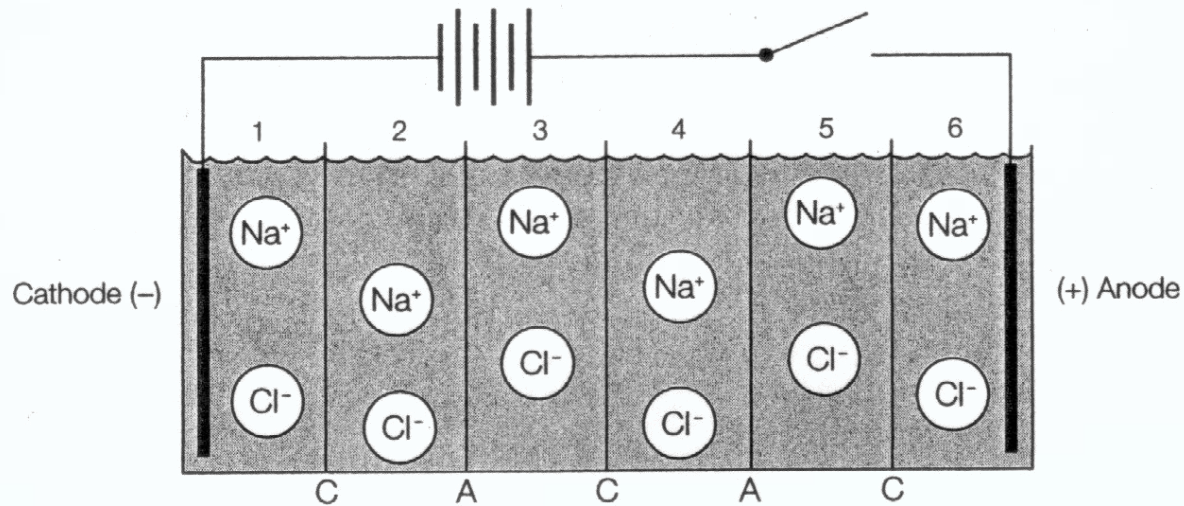
Electrochemistry



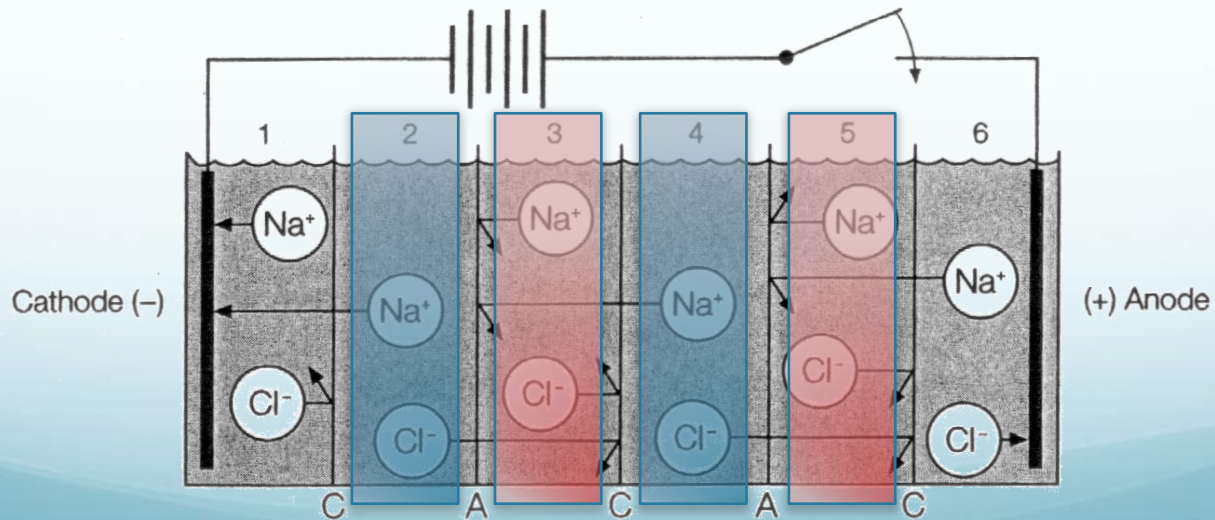
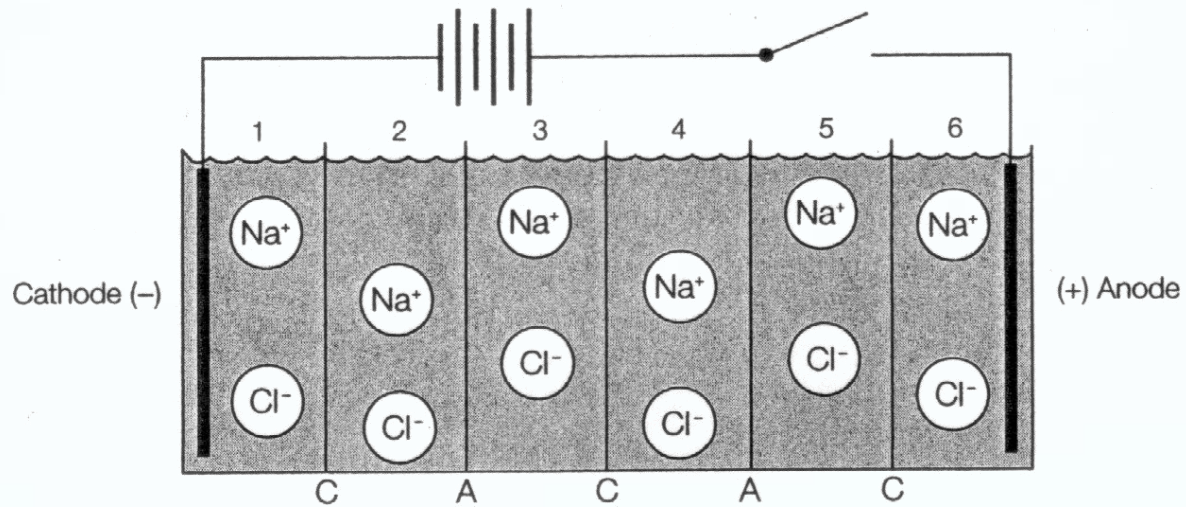
Principles of Electrodialysis



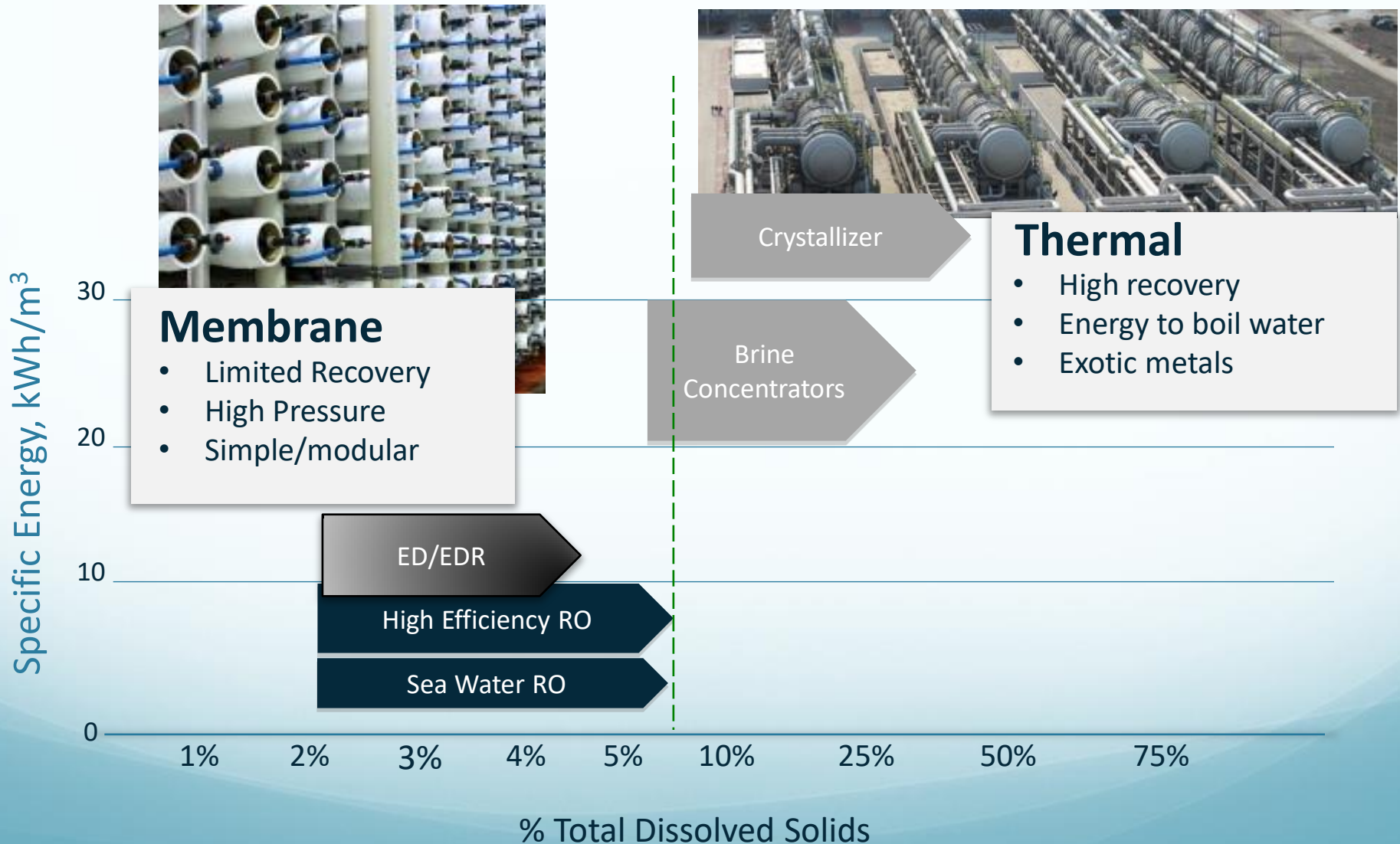
Principles of Electrodialysis



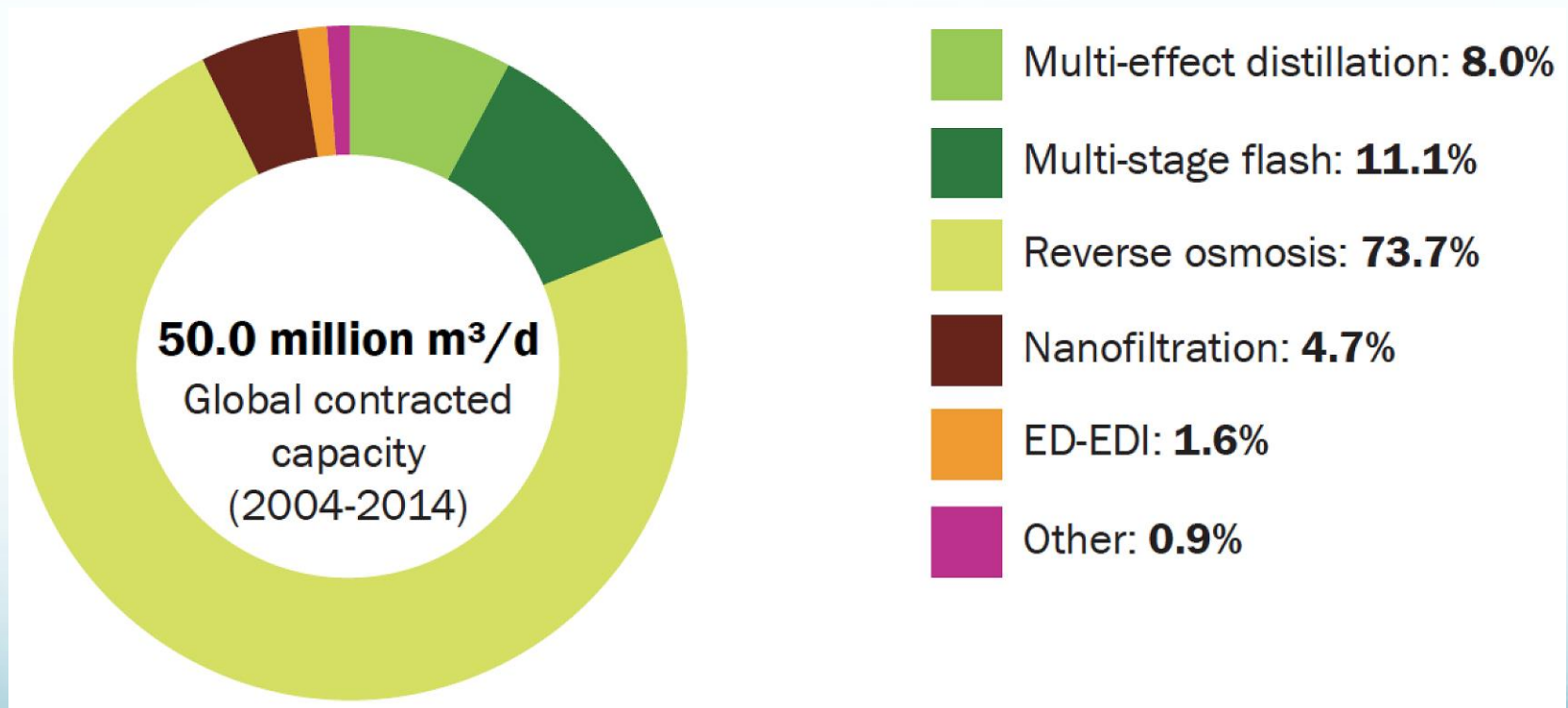
Principles of Electrodialysis



Desalination Technologies

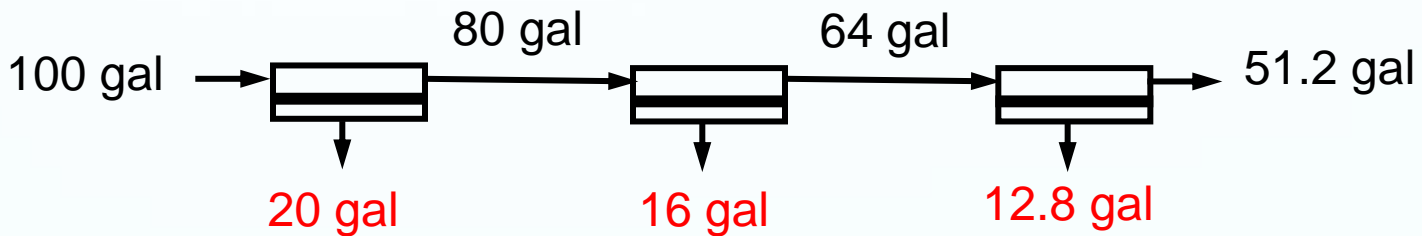


Contracted Capacity by Technology



Water Recovery in RO/NF Systems

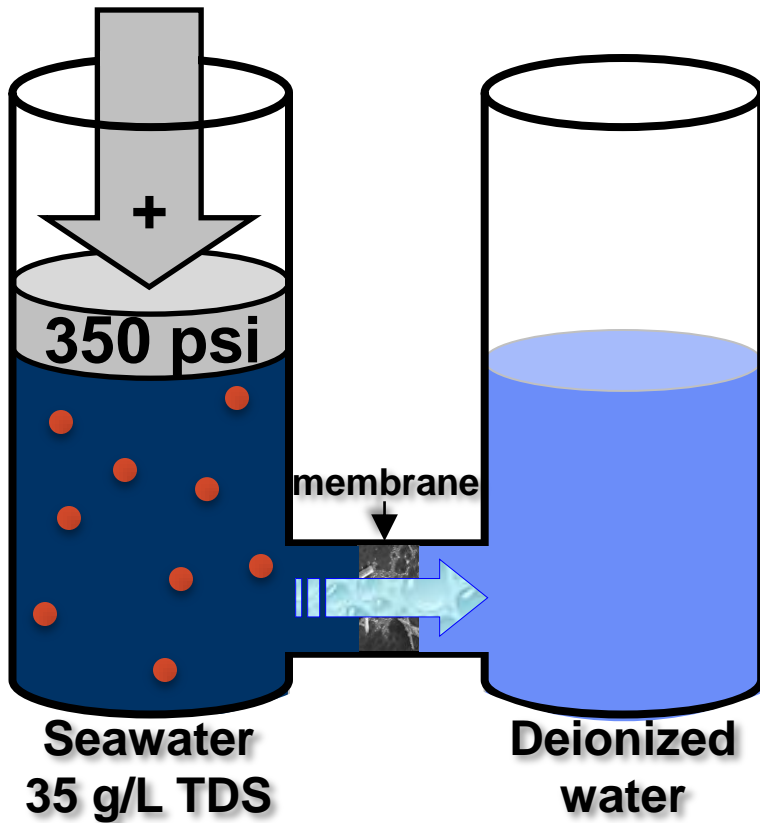
- Assuming each membrane (or each stage) operates at 20% recovery



- Total permeate from all membranes (stages) = 48.8 gal

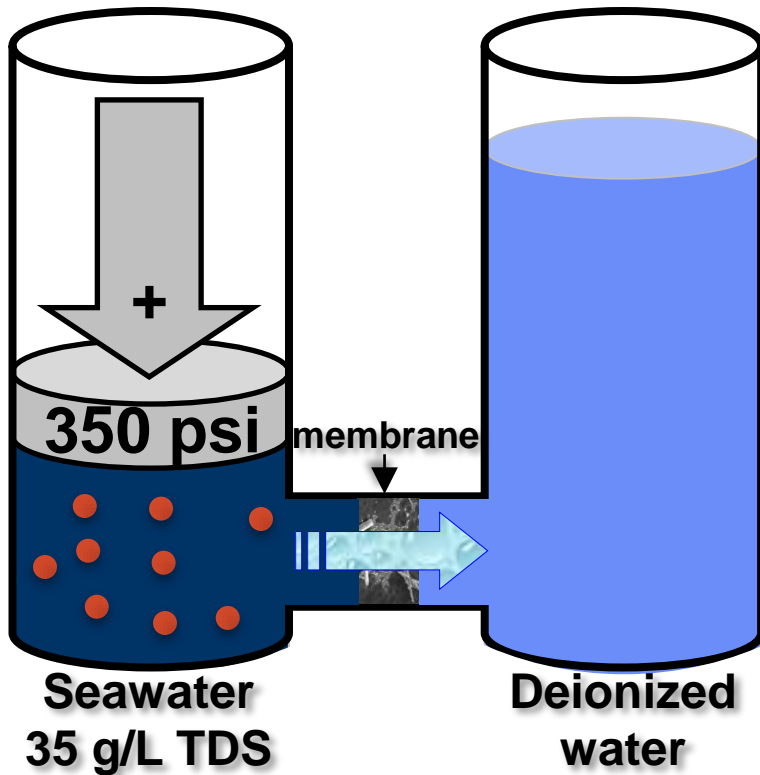
$$\text{Overall Recovery} = 100 \left(\frac{48.8 \text{ gal}}{100 \text{ gal}} \right) = 48.8\%$$

Overcoming Osmotic Pressure



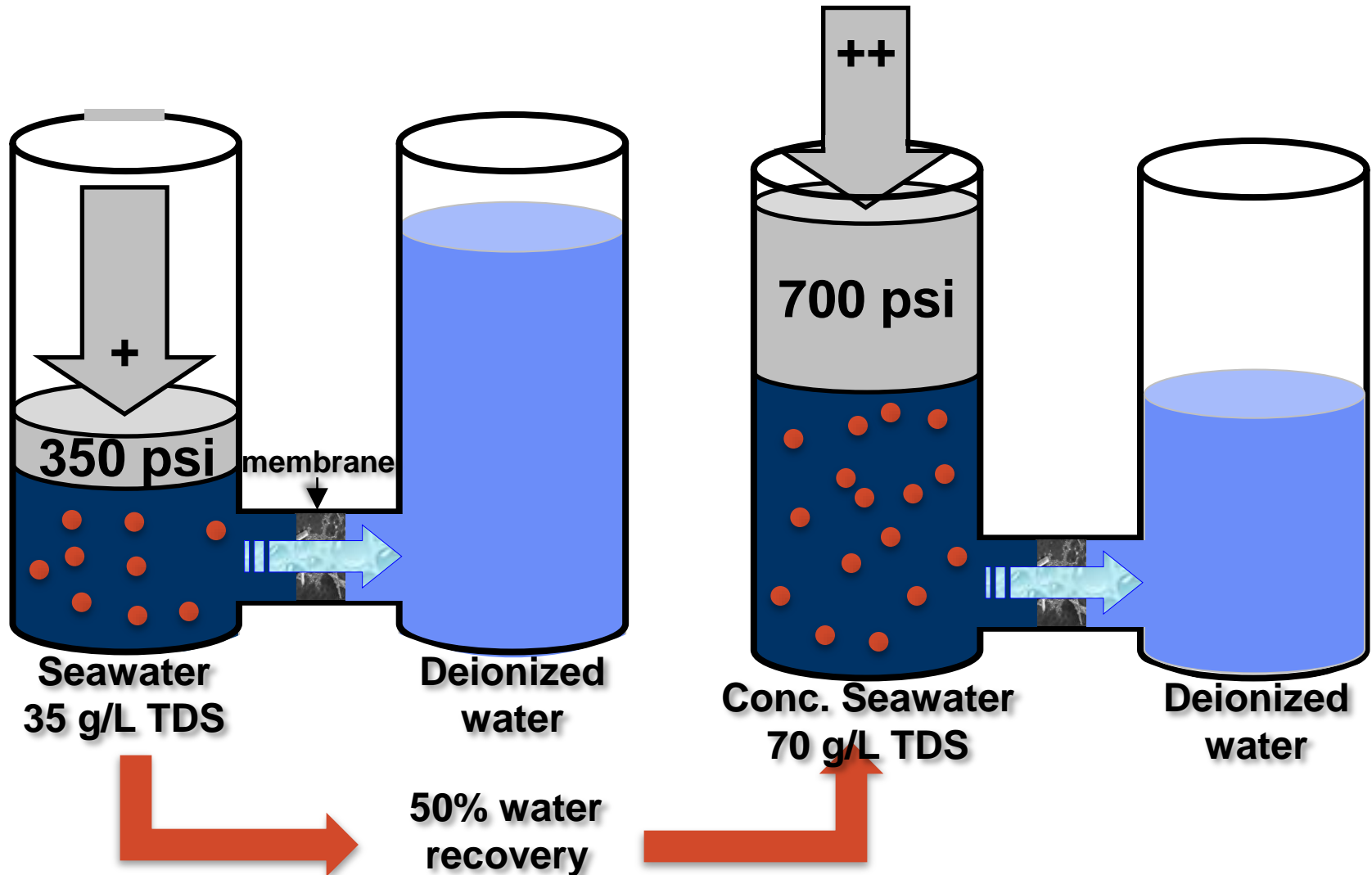
Overcoming Osmotic Pressure

Rule of thumb: 1 g/L salt \approx 10 psi of osmotic pressure



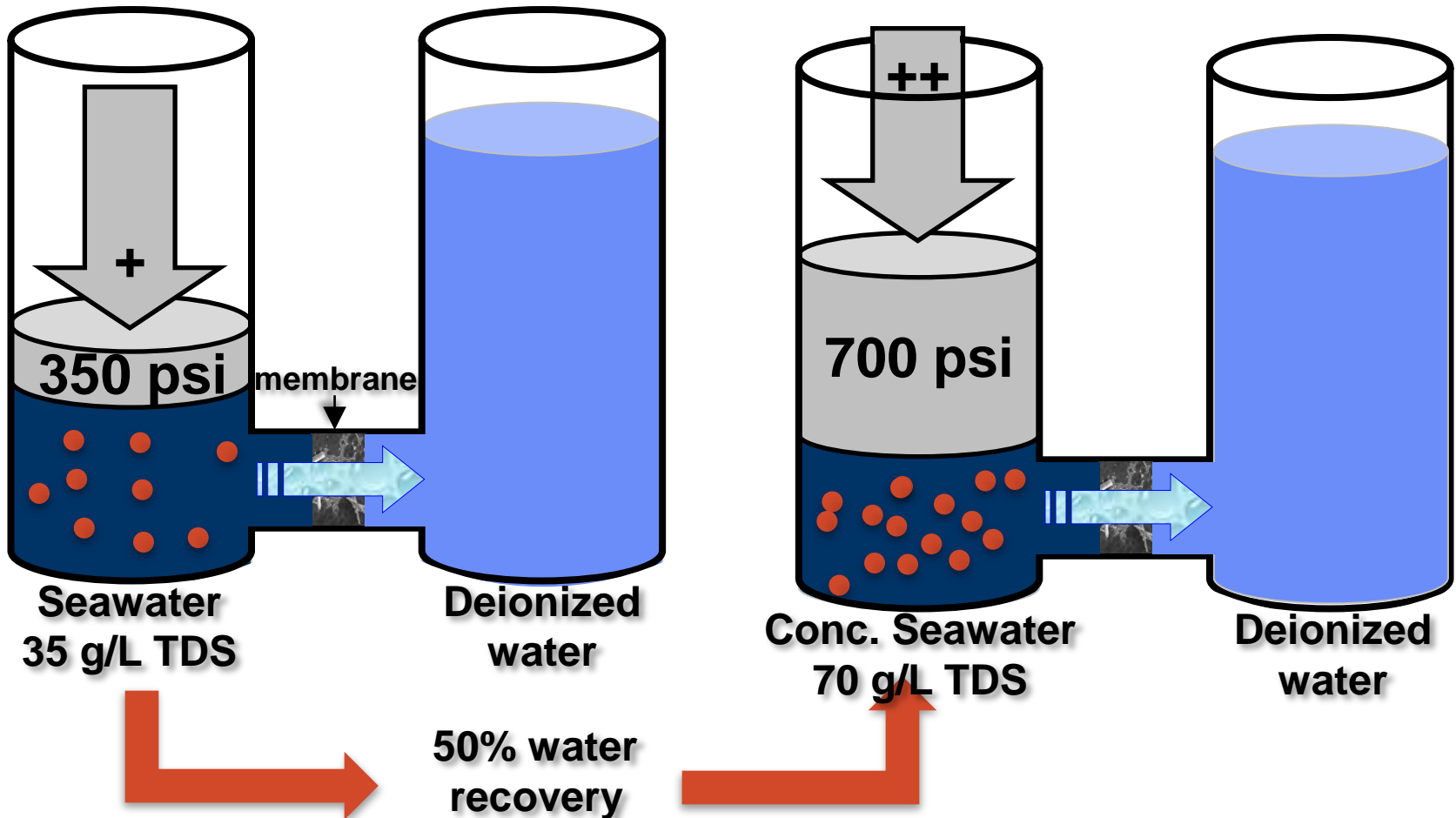
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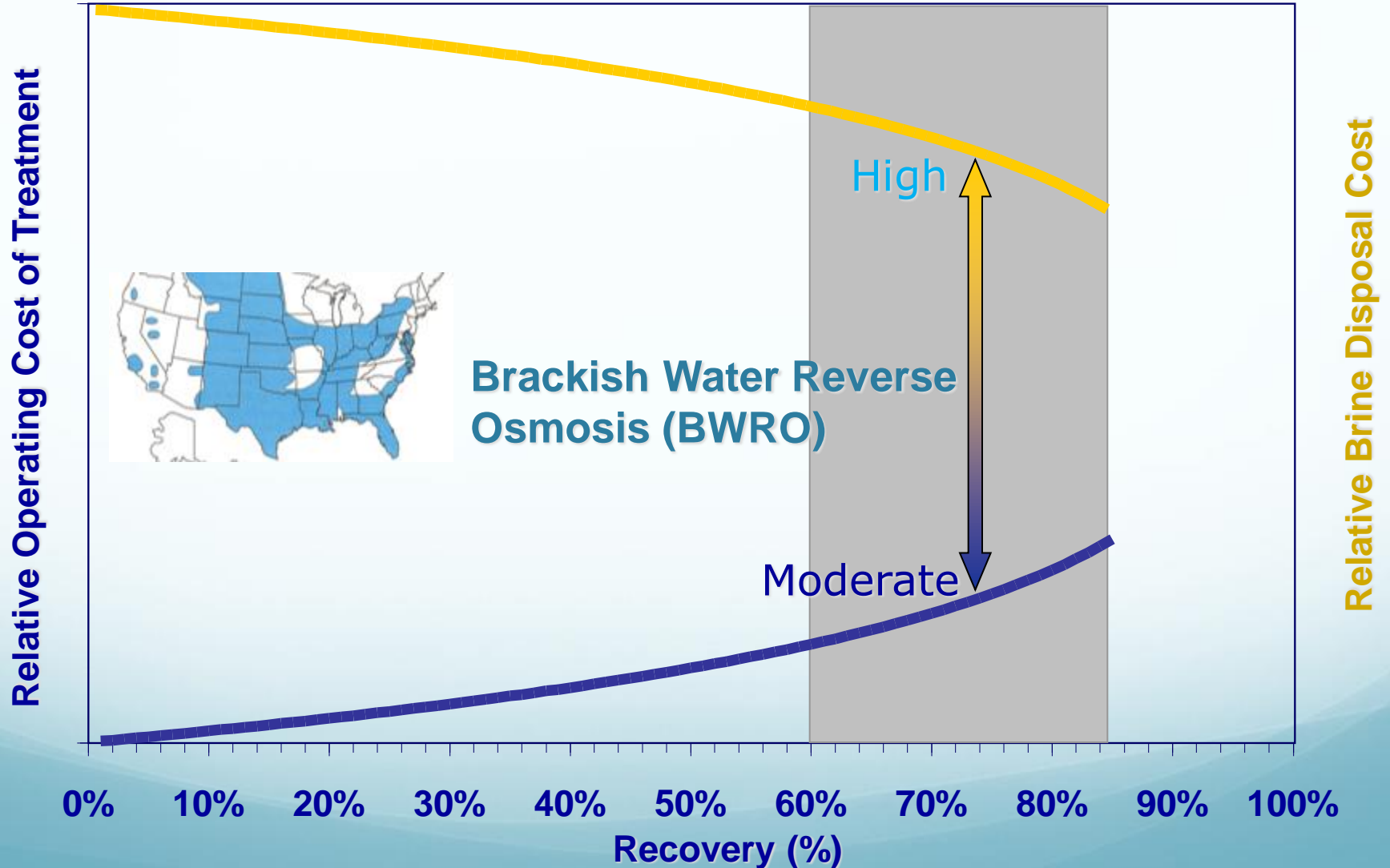


Overcoming Osmotic Pressure

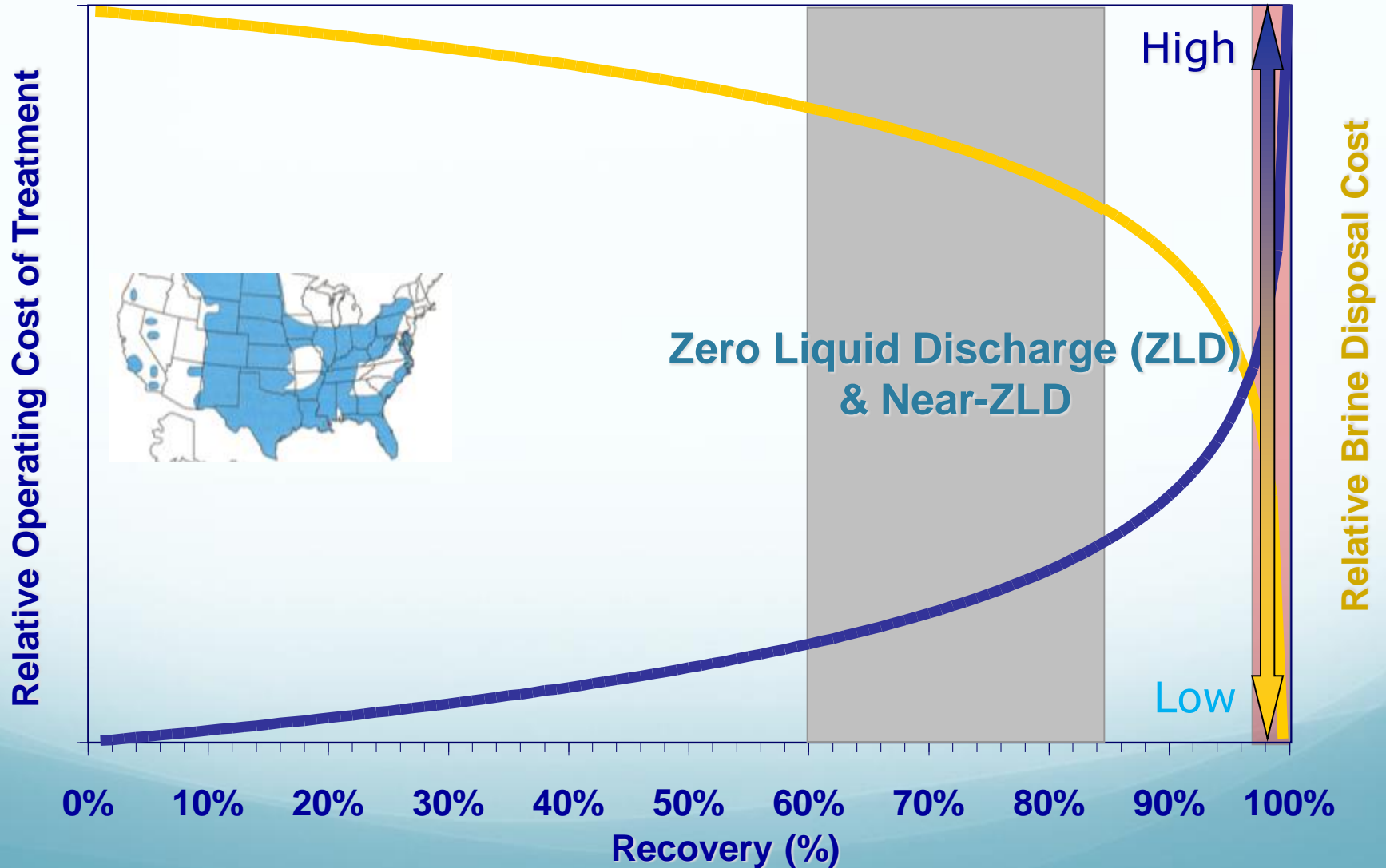
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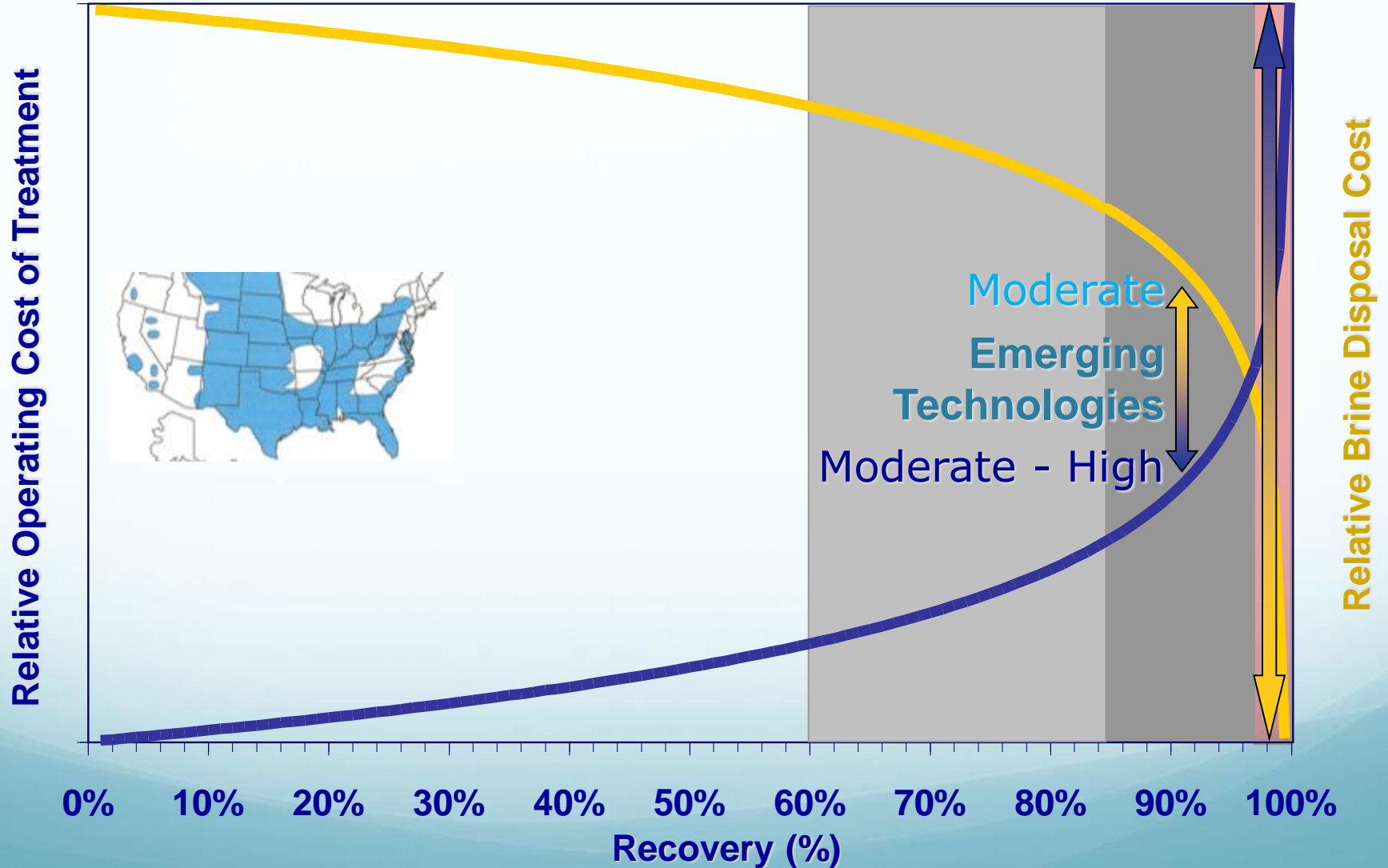
Water Recovery and Associated Costs



Water Recovery and Associated Costs



Water Recovery and Associated Costs

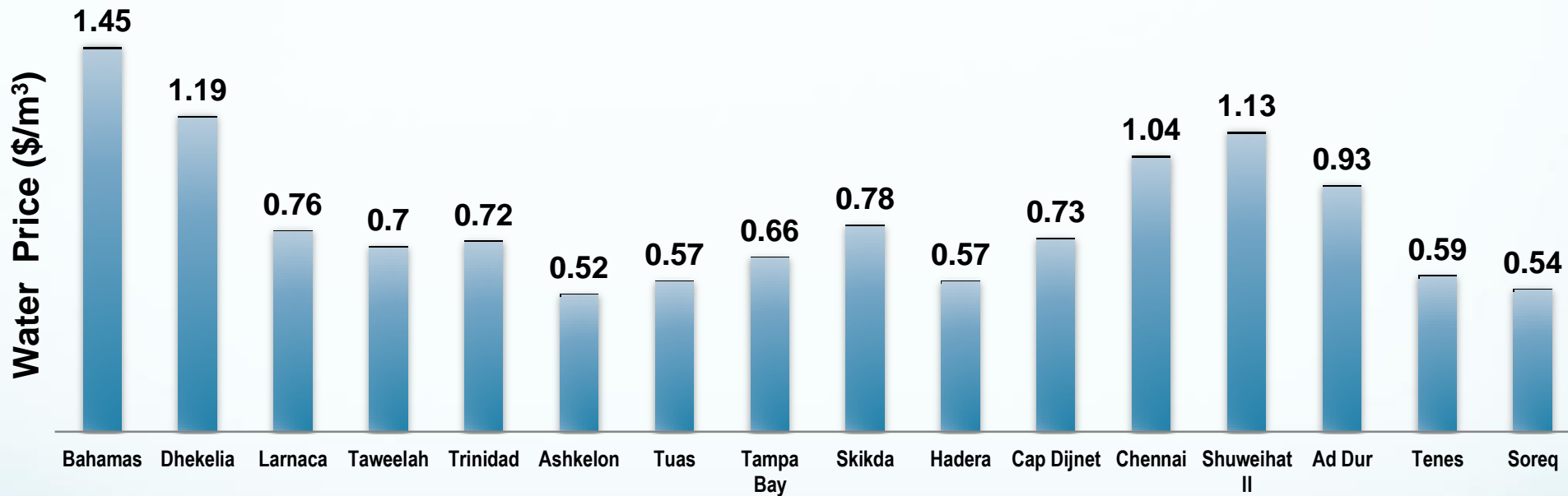


Relative Desalination Costs

Process	Capital Cost \$/GPD	O&M Cost \$/kgal	Water Cost \$/kgal (\$/m ³)
Nanofiltration	1.00 – 1.30	0.40 – 0.70	0.60 – 1.00 (0.16 – 0.26)
Brackish water Reverse Osmosis	1.20 – 2.50	0.80 – 1.50	1.50 – 3.00 (0.40 – 0.80)
Seawater Reverse Osmosis	3.50 – 5.00	2.00 – 4.00	2.50 – 7.00 (0.66 – 1.85)

GPD – gallon/day
kgal – 1,000 gallon

Seawater Reverse Osmosis Desalination Plants



Cost difference:

- Energy cost
- Concentrate management
- Project cost (engineering, permitting, etc.)
- Tax credits and other financial costs/incentives
- Size

Pre-treatment

Removal of:

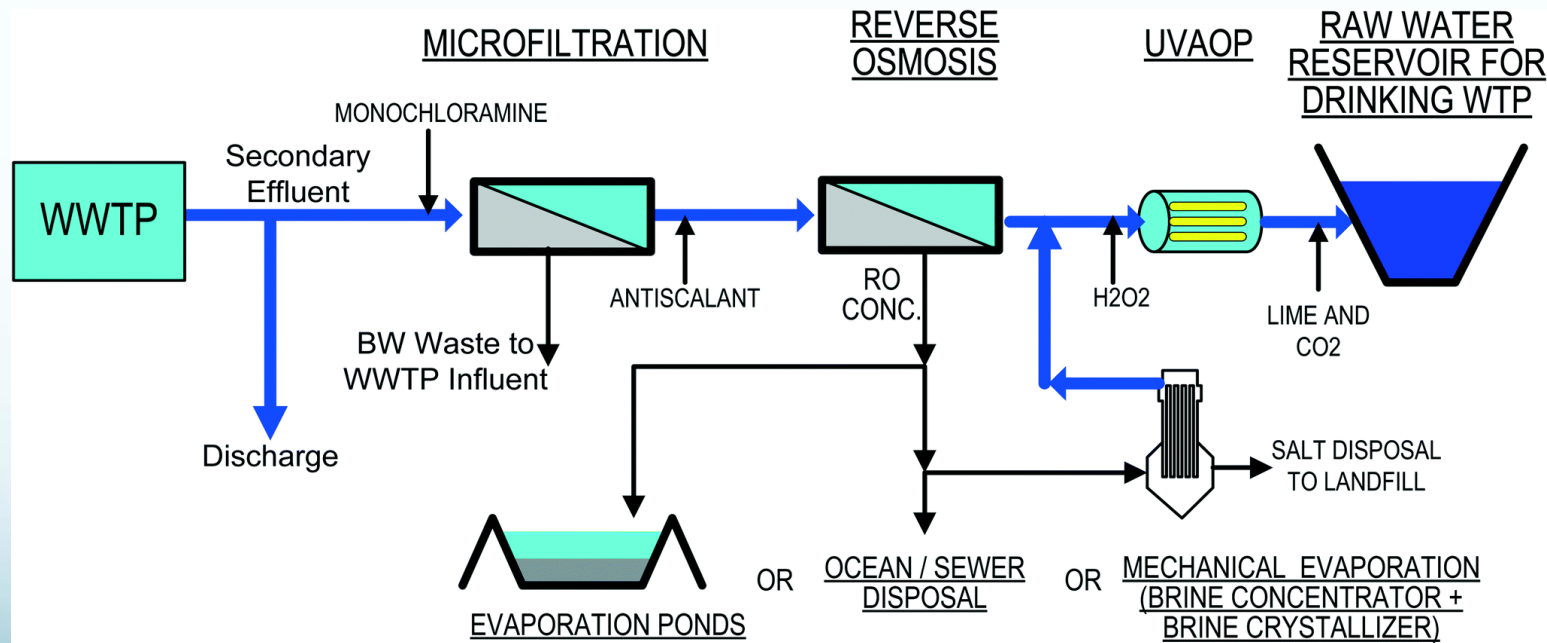
Suspended solids or emulsions

Microorganisms

Organic matter

Specific ions (iron, calcium, magnesium, silica, etc.)

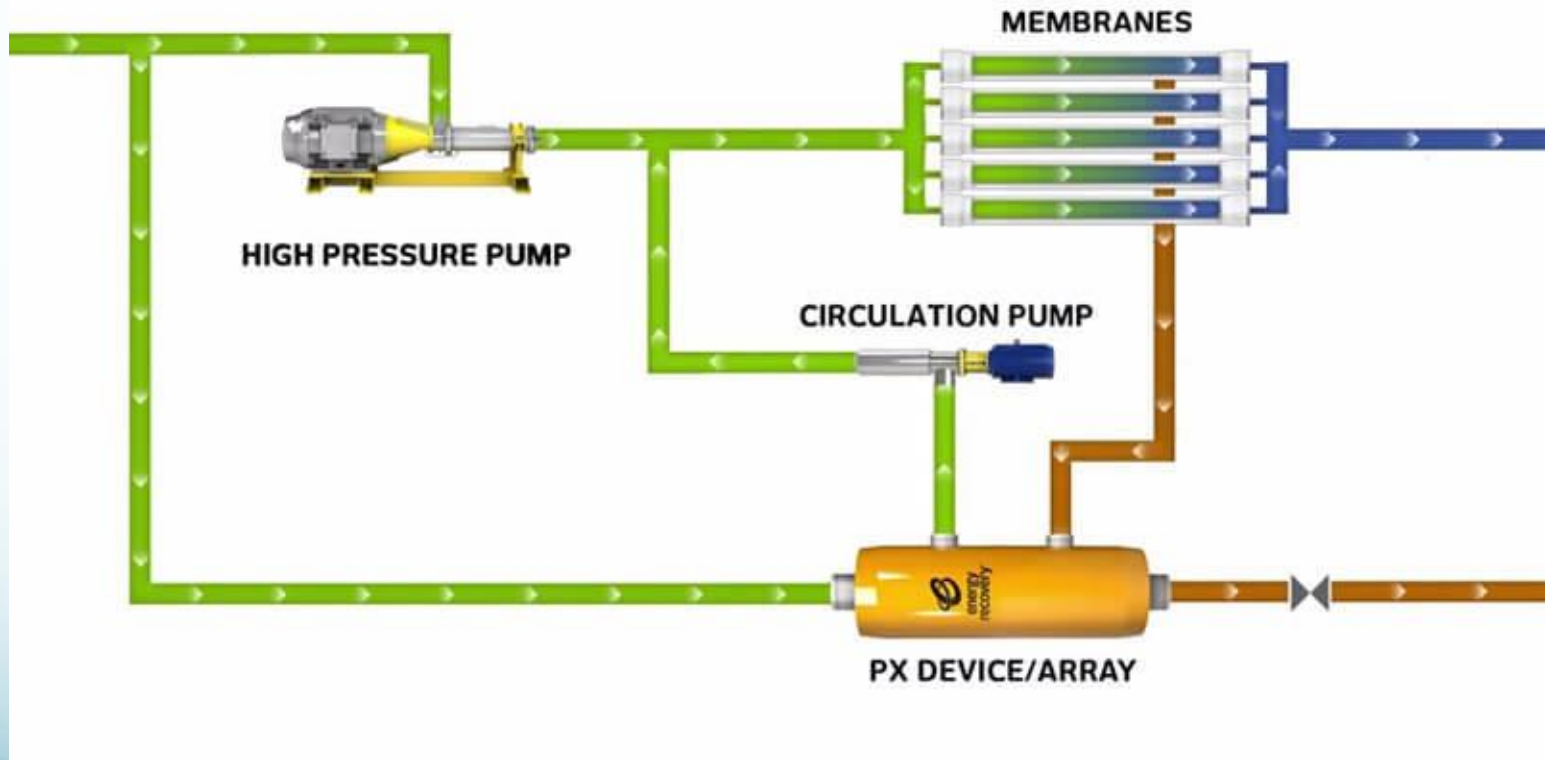
Pre-treatment and Desalination for Potable Reuse



Energy Recovery

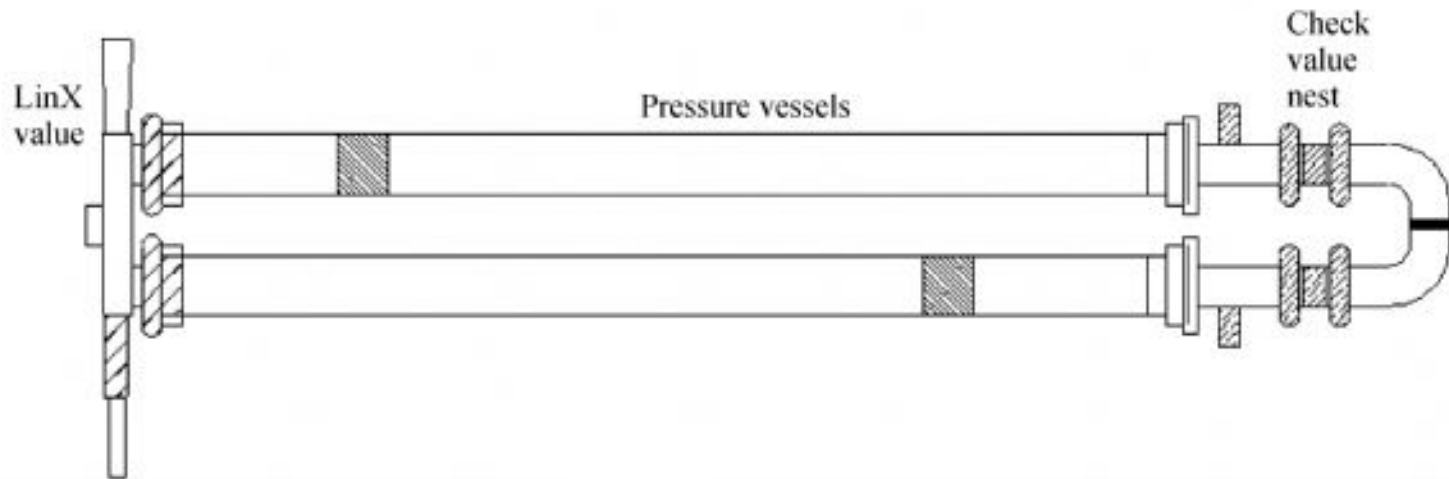
Pressure Exchangers

ERI (Energy Recovery Inc.)

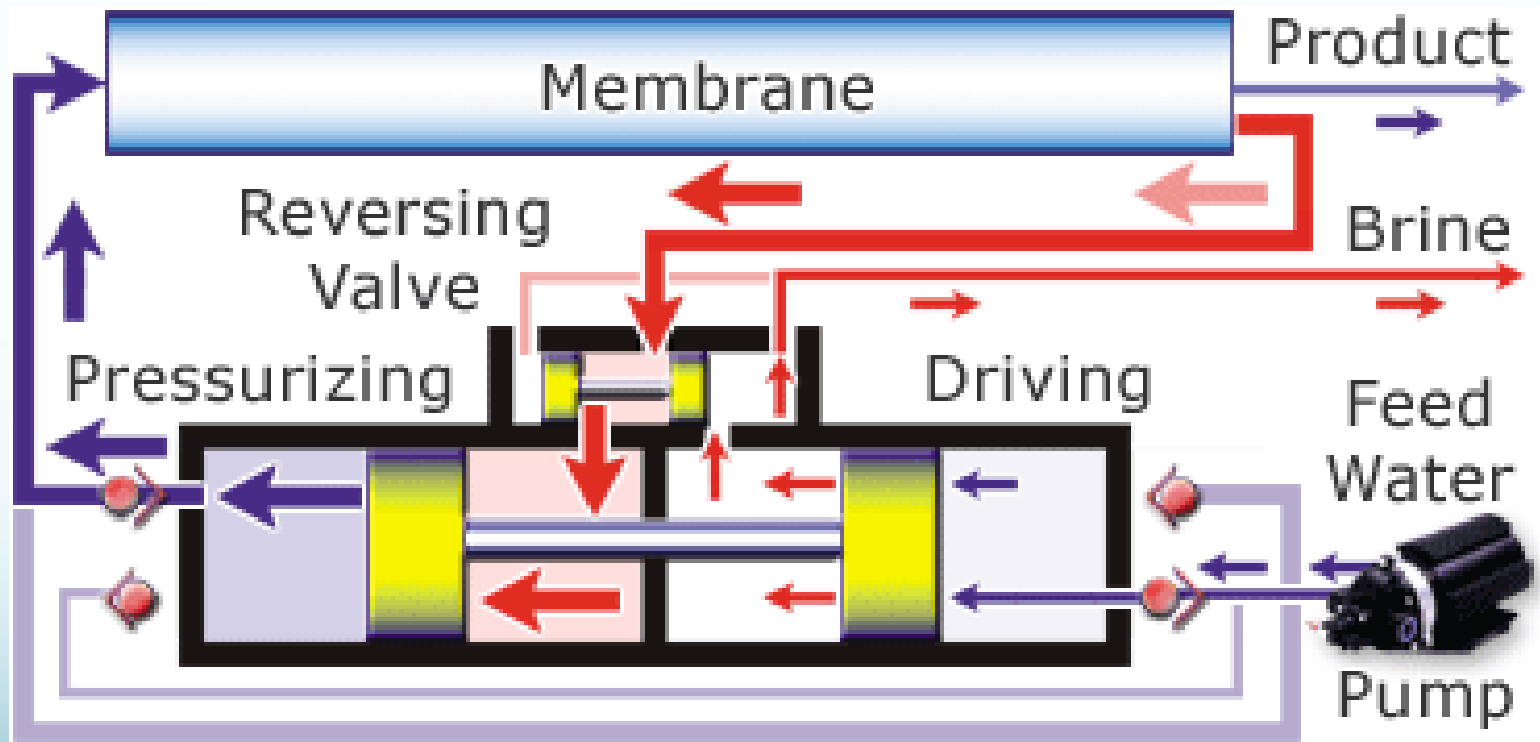


Pressure Exchangers

DWEER (Dual Work Exchange Energy Recovery)

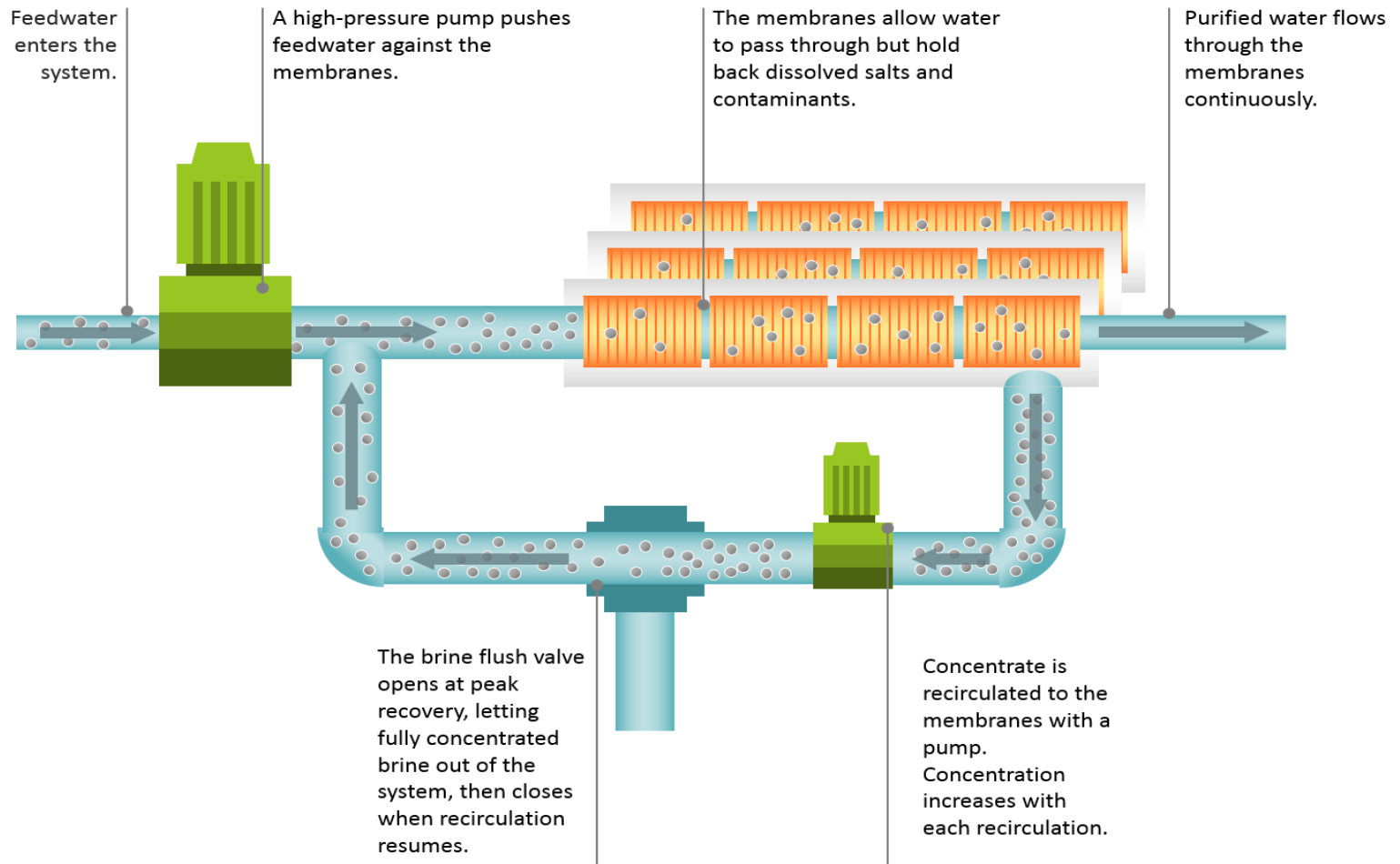


Clark Pump

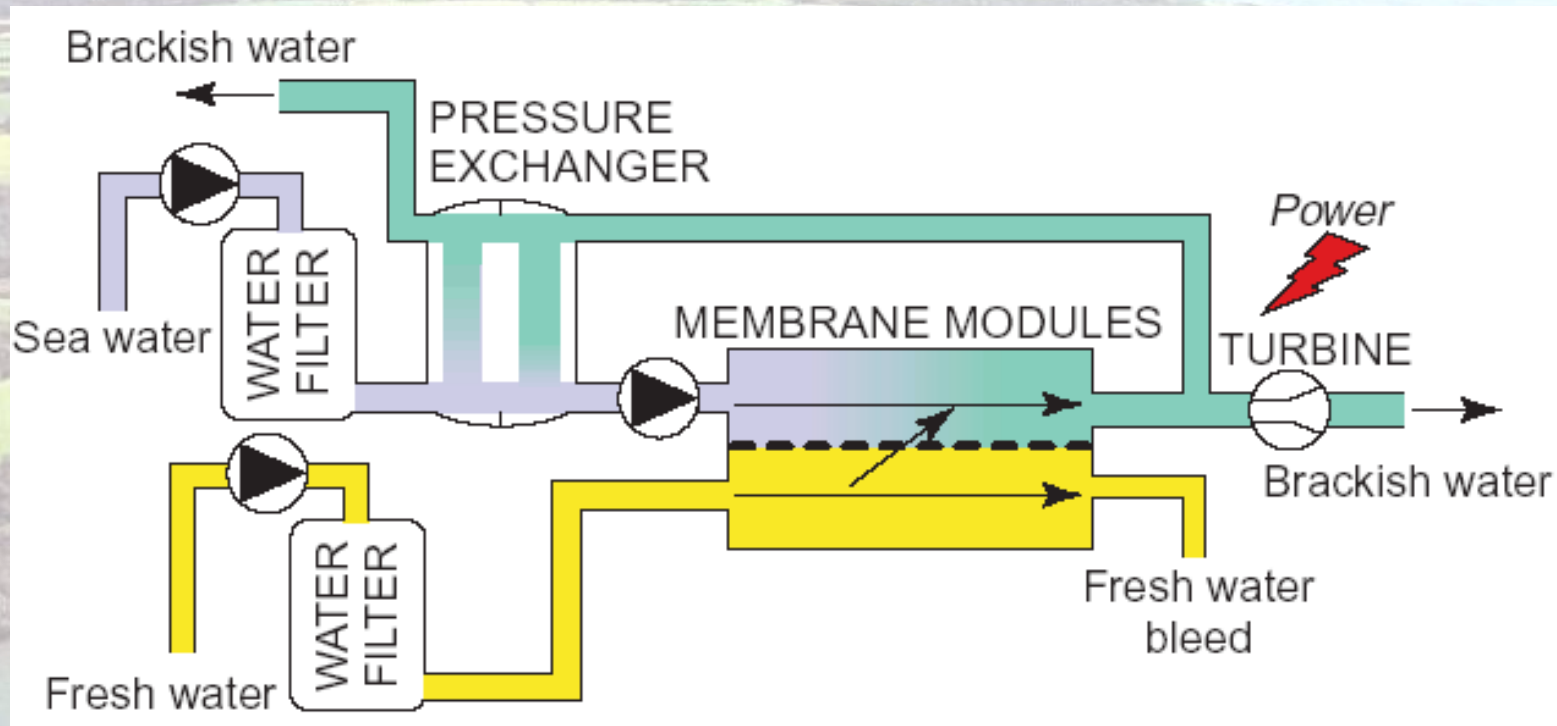


Desalitech CCD

(Closed Circuit Desalination)



Osmotic Power Energy Recovery: Pressure Retarded Osmosis

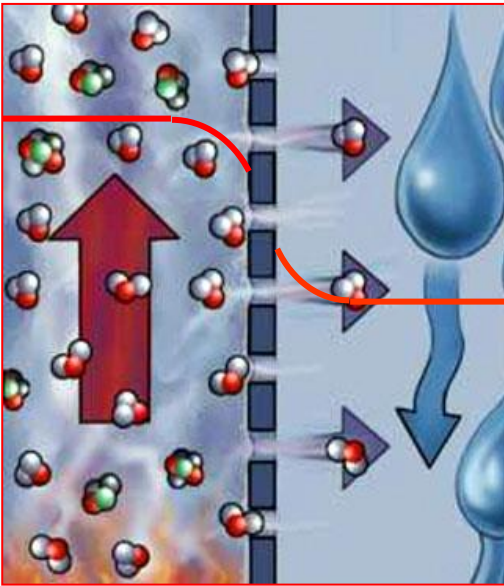


Concentrate Management

- Coastal
 - Ocean outfall
- Inland
 - Surface discharge (bad!)
 - Sewer disposal (not good!)
 - Deep well injection
 - Land application/dust control
 - Zero liquid discharge (ZLD) (crystallization, expensive...)
 - Pipeline to ocean (not common)

Future Technologies (?)

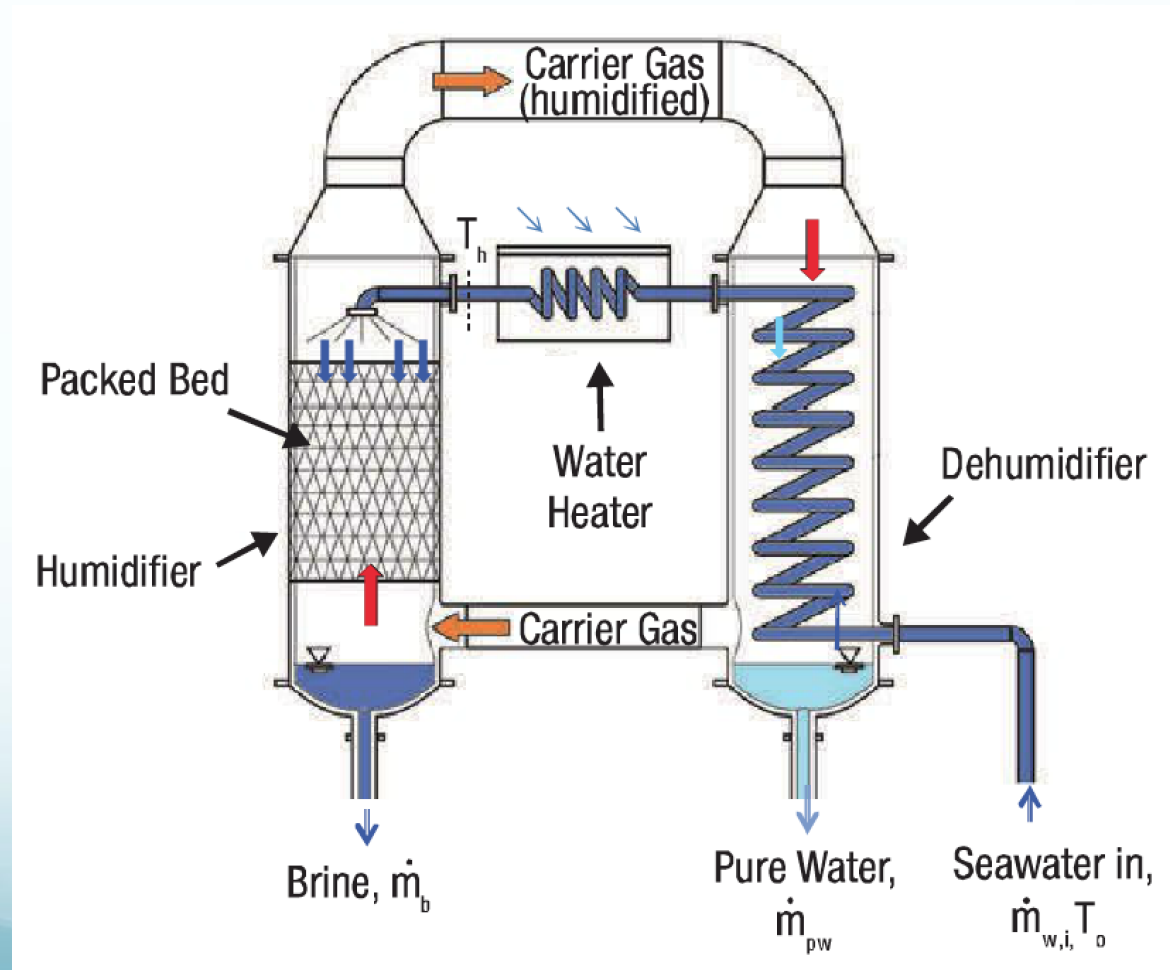
Other Desalination Technologies: Membrane Distillation



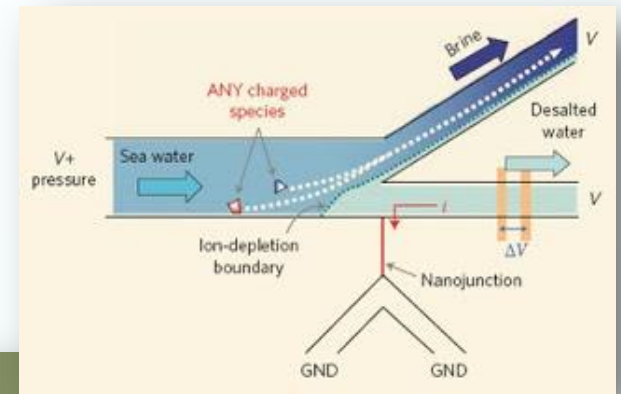
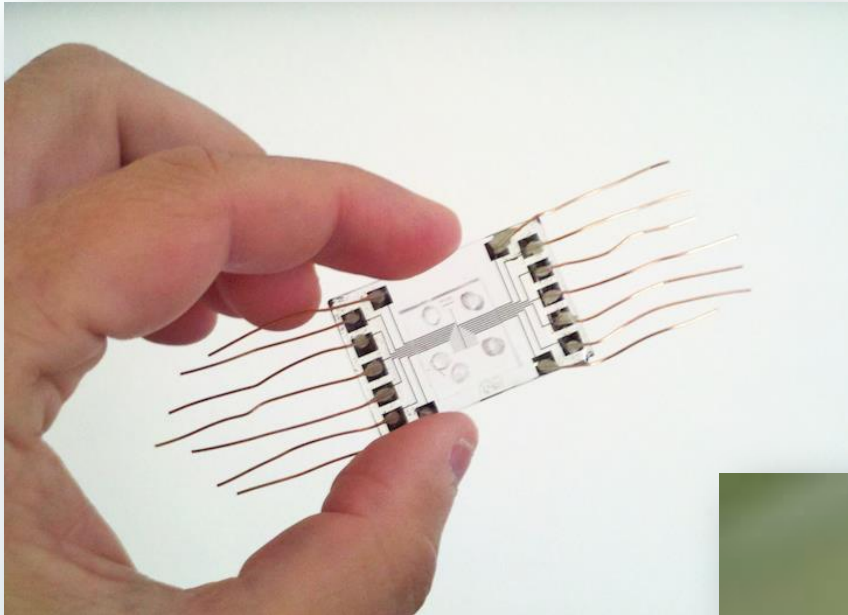
adapted from: <http://www.water-technology.net/>

- Heated aqueous feed solution is brought into contact with the feed side of a hydrophobic, microporous membrane
- Colder water or gas is in contact with the distillate side of the membrane
- Vapors diffuse through the pores and condense in the colder stream

Other Desalination Technologies: Humidification-Dehumidification



Other Desalination Technologies: “water chip”



In summary...

- Desalination is used for salt removal, but also for purification of water and removal of other contaminants
- Development of new membranes focuses on increasing water flux, increasing solute rejection, and increasing chemical resistance of membranes
- Reverse osmosis is a core process to enable potable water reuse
- Desalination (especially reverse osmosis) is affordable and provides a reliable source of potable water; and efficient energy recovery makes it possible
- Brine management is a prevailing problem, especially at inland installations, and it drives development of high-recovery desalination (ZLD)

Thank you

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CEE URL: <http://cee.mines.edu/>  
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